# Borehole Mining of Manganese at Emily, MN

Or...Mn in MN by Michael Liljegren, DNR

#### This talk will cover:

Geology of the Manganese Deposit in Emily

- Proposed Mining Methods
- > Bulk Sample Collection Project
  - Permitting Requirements
  - Hydrogeologic Evaluation

#### **Overview of the Project**

Collect a "bulk sample" (>12,000 cubic yards) of manganese-rich iron formation from 1 borehole > Evaluate borehole mining technology • Uses high-pressure water jet to mine in situ

- Ore-water slurry is pumped out, filtered and reinjected
- Filtered material (ore) is trucked to U of M minerals research facility in Coleraine for processing
- EAW developed ONLY for collecting the bulk sample EIS will be required if full-scale operation is implemented

#### Who is proposing to do the project?

- Cooperative Minerals Resources (CMR) a wholey-owned subsidiary of Crow Wing Power Cooperative
  - Profits will be shared with their 36,000 members – mostly in Crow Wing County
- Environmental permitting & hydrologic evaluations by Barr Engineering Co.
- > Bulk sample collection to begin in August 2010

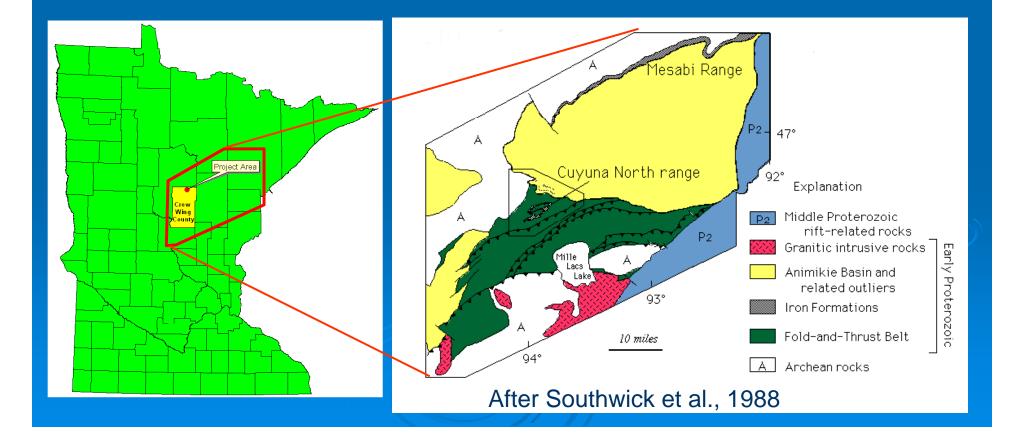
#### Geology of the Manganese Deposit at Emily, MN



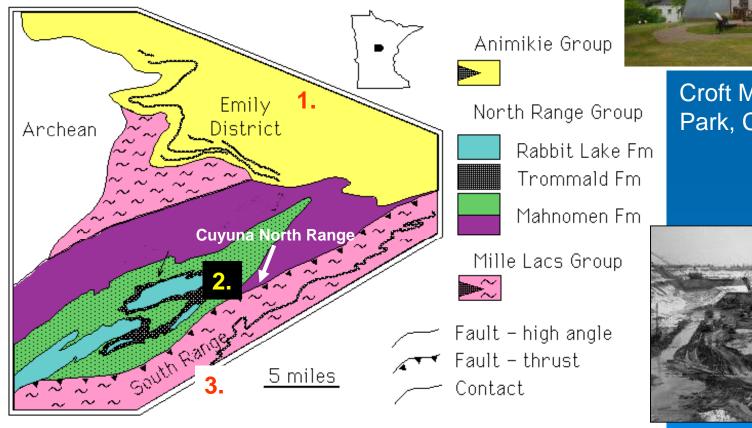
Photo of Borehole Mining Site (by Ellen Considine, Barr Engineering Co.)

#### The Cuyuna or "Old" Iron Range

Iron mining from 1904 to 1984
 106 million tons for iron ore mined



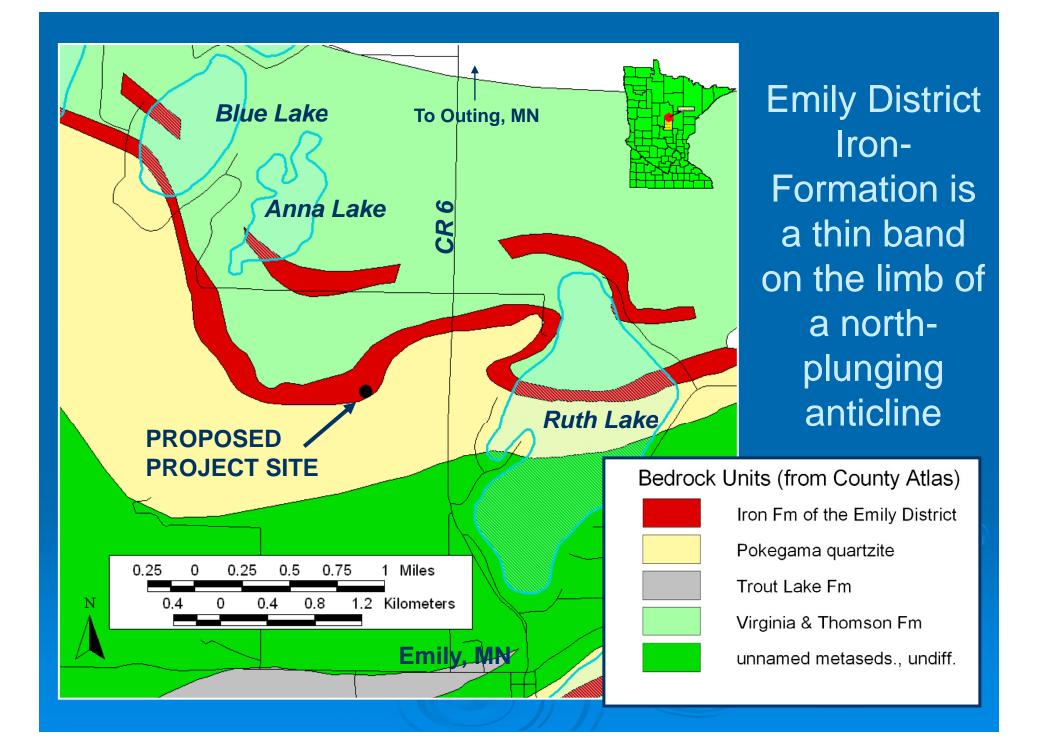
#### Cuyuna Range is Divided Into 3 Districts Emily District > North Range South Range

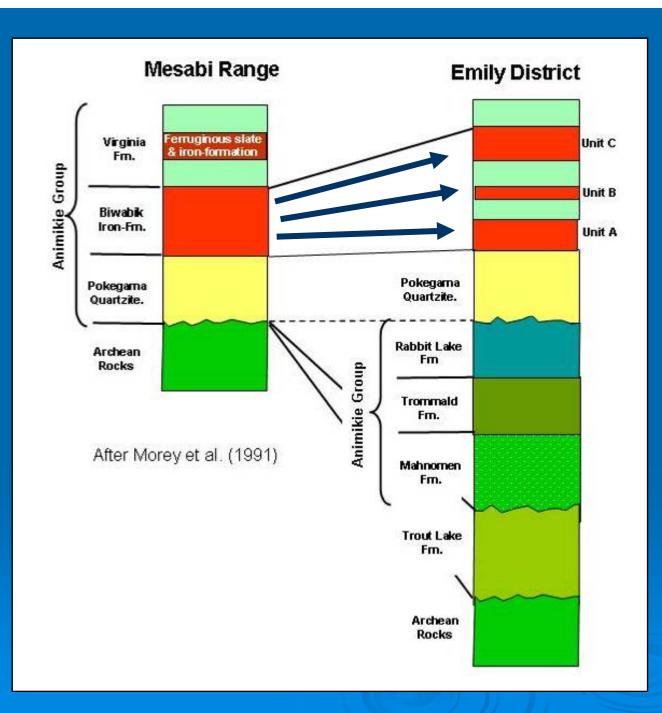




#### **Croft Mine Historical** Park, Crosby, MN





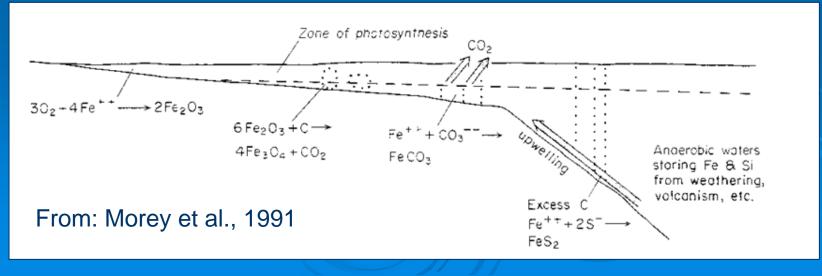


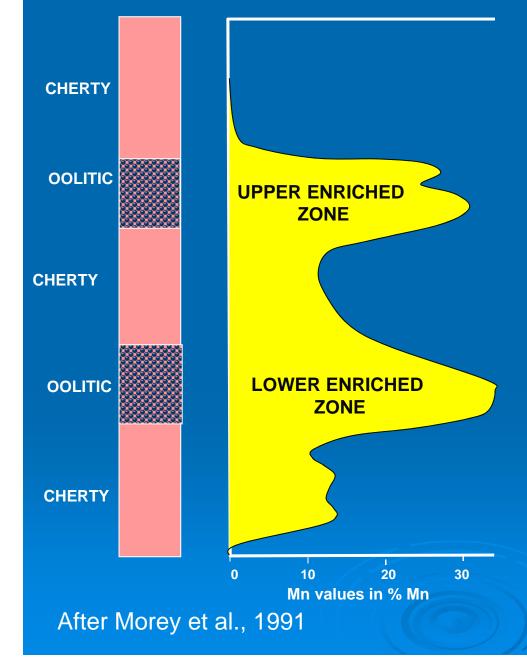
Emily **District Iron-**Formation Units A, B, & C correlate with the **Biwabik** Iron-Formation of the Mesabi Range

#### **Depositional History**

Similar to Biwabik Iron Formation – chemical precipitation of iron and silica in a shallow Precambrian sea

> Abundance of oolitic hematite indictes wave-action reworking





Manganese Enrichment is Present in Two Zones Corresponding to Oolitic Zones

#### **Reflux Model of Enrichment**

Manganese and barium were carried to their final depositional site by anaerobic water systems.

Both precipitated when the anaerobic water met and mixed with aerated water in uncemented iron-formation on the seafloor.

#### Manganese Mineralogy of the Deposit

Manganite - MnO(OH) – primary ore mineral
 Psilomelane – (Ba,Mn)<sub>3</sub>(O, OH)<sub>6</sub>Mn<sub>8</sub>O<sub>16</sub>
 Cryptomelane - K(Mn)<sub>8</sub>O<sub>16</sub>

 Manganese Concentration (by weight) greater than 50% in 2 zones (i.e. "high-grade" ore)
 Hematite is dominant iron mineral
 Almost <u>zero</u> sulfur

#### What is Manganese Used For?

- Essential to iron and steel production by virtue of its sulfur-fixing, deoxidizing, and alloying properties (there is no substitute)
- Alloying agent in aluminum (especially in beer cans really!)
- New generation batteries
- Pollutant removal from coal-fired power plant emissions (Pahlman process)

#### World Sources of Manganese

#### > High-grade (> 44% Mn)

- 680 million tons ore world-wide
- Mostly in southern hemisphere countries using for internal use (limited export)
- Low-grade (<44% Mn)</p>
  - Russia's low-grade ores are depleting
  - China has very low-grade ore for internal use
  - Thin layers of 25% Mn in nodules on ocean floor (not yet mined)

Total world production and consumption of Mn ore in 2003 was 23 million tons

#### U.S. Manganese Sources

- The US imports <u>ALL</u> its manganese from Gabon, South Africa, France, and Brazil (692,000 tons in 2003)
- The US currently has no high-grade (>44%) Mn reserves
- Strategic stockpiles of Mn in the U.S. are essentially depleted
- Manganese from recyled materials is negligible
- Recent price of Mn: @ \$1.30/lb

# How much Manganese is Available at Emily?

Bureau of Mines estimated @ 2 billion pounds (1 million tons) in a 9 acre area within 2 zones (50 feet and 70 feet thick) (Pahlman, 1995)

The deepest of these zones is @ 400 feet below ground surface

Likely the largest high-grade Manganese deposit in the Northern Hemisphere.

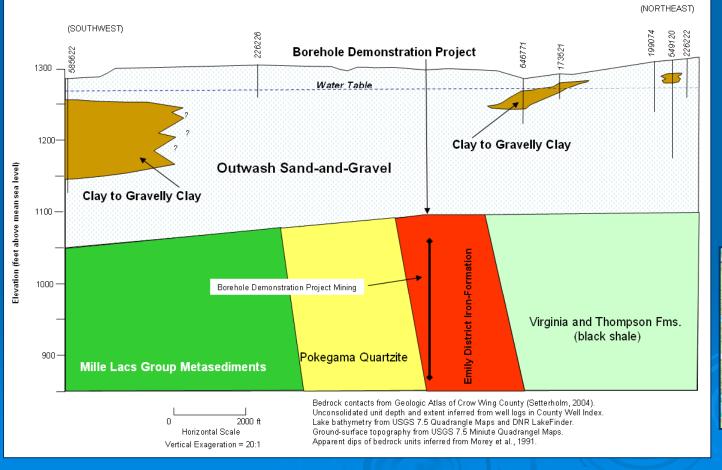
### Mn-Rich Sections of Iron-Formation are Very Friable

This is very important in determining whether or not this formation can be mined in an environmentally friendly manner

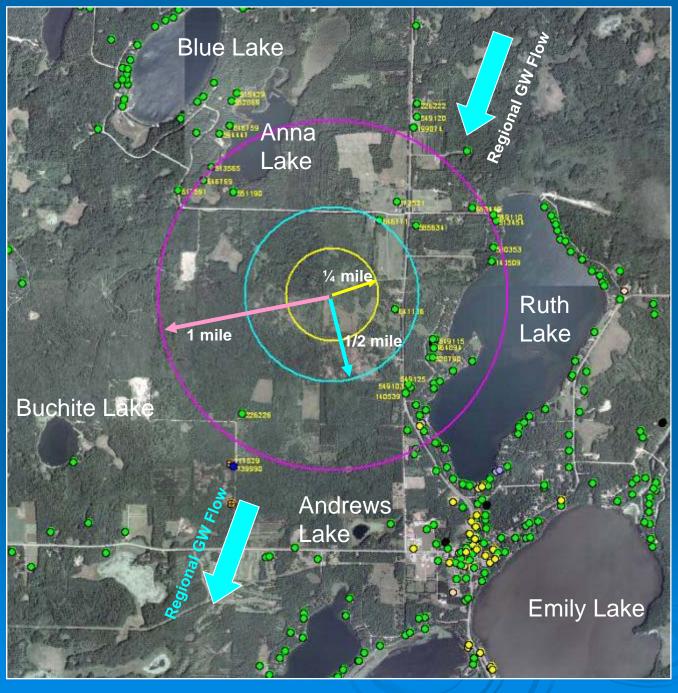


Rotosonic core of iron-formation

#### Bedrock is Overlain by @ 180 – 200 Feet of Sand-and-Gravel Outwash







Depth to Groundwater is @ 35 Feet at the Project Site

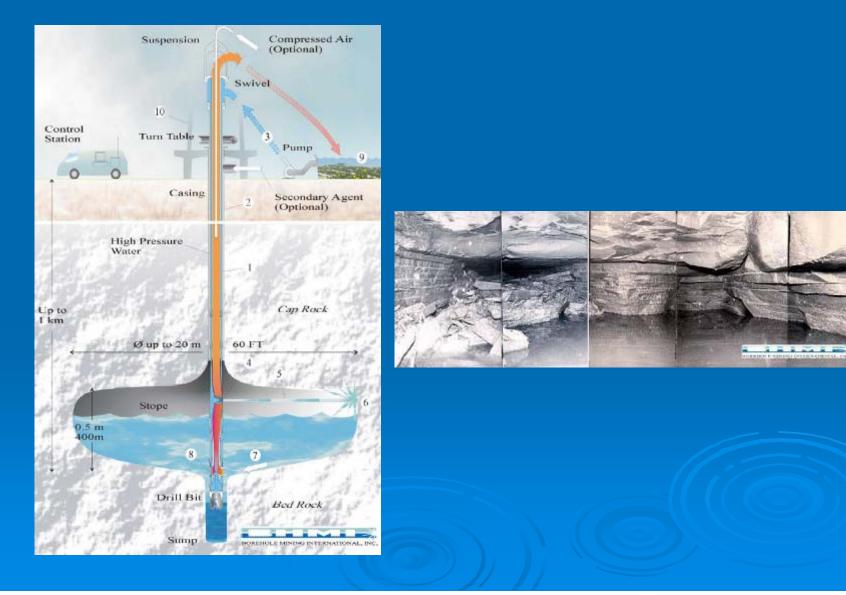
The soils are VERY sandy.

Lakes are in direct hydraulic connection with the sand-andgravel aquifer.

All wells in the area completed in sandand-gravel aquifer

Dots show CWI well locations

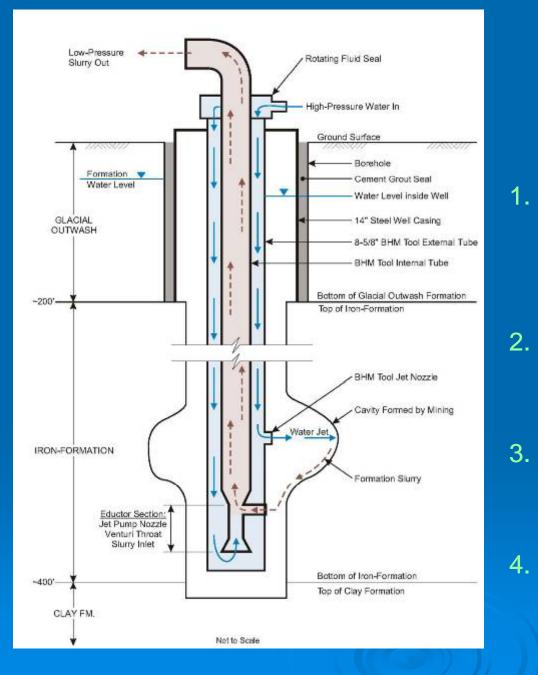
#### The Borehole Mining Process



#### **Possible Mining Approaches**

- Open-Pit mining: Have to deal with 200 feet of saturated overburden and dewatering.
- Underground mining: Likely extensive dewatering and expensive.

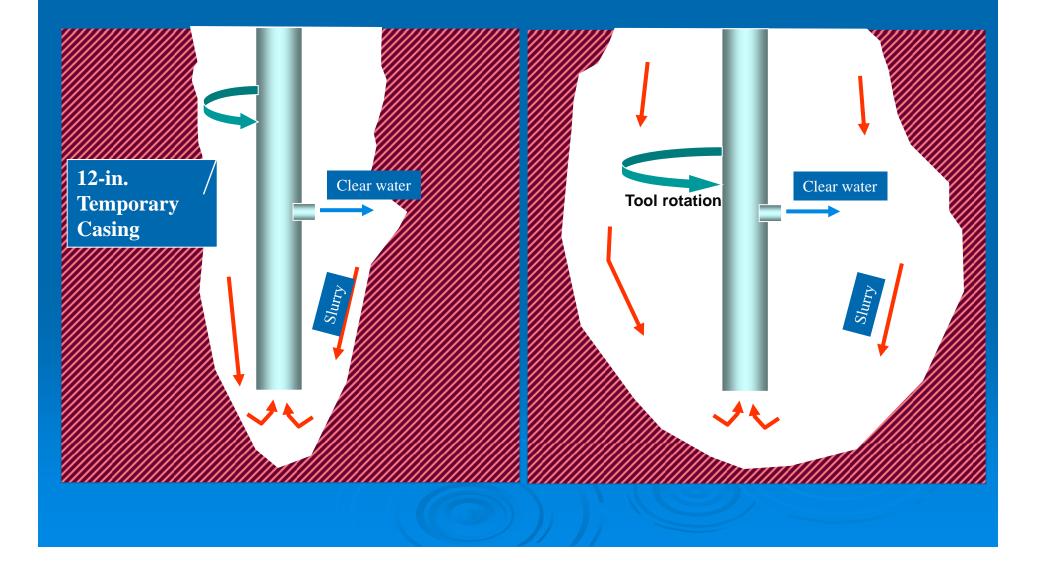
In situ leach mining: Studied by Bureau of Mines and deemed practical – but probably environmentally unacceptable



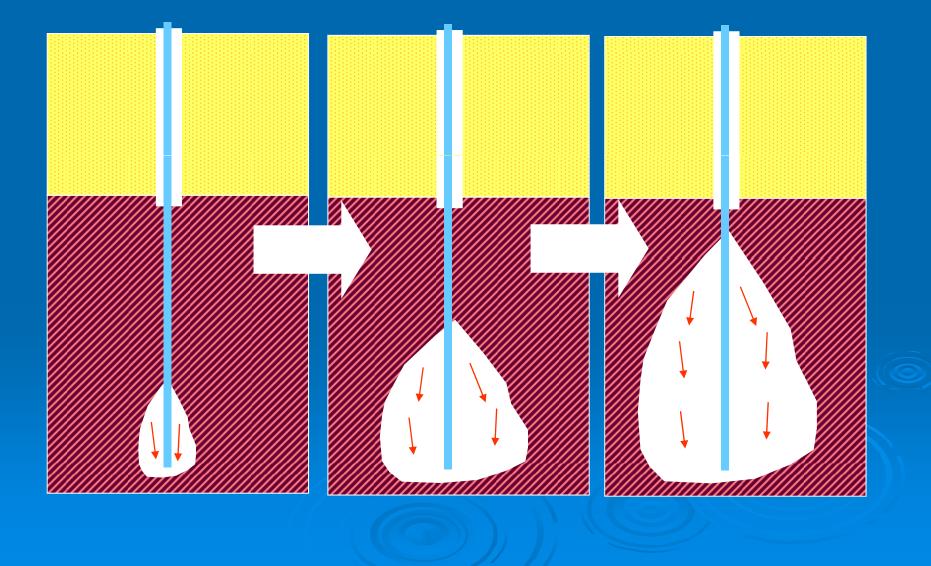
Borehole Mining is performed hydraulically

- Water is jetted into the formation at @ 1,200 – 1,800 psi. The tool head rotates.
- 2. Sloughed deposit settles to the bottom of the borehole.
- 3. The slurry is pumped out, screened, and filtered at the surface.
  - Clear water is reinjected as water jet

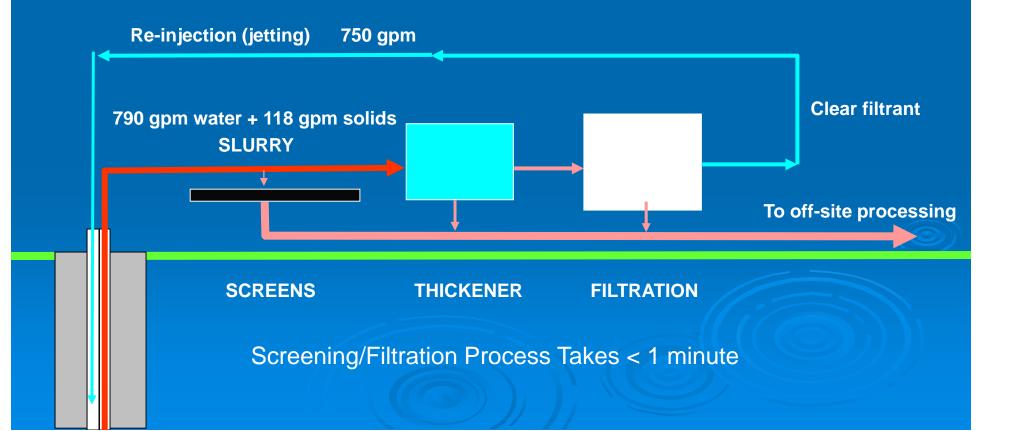
## Water Cycling in Borehole



### **Borehole Mining Progression**



### General Filtration Process at Ground Surface



#### Expected Operation and Water Pumping

Net instaneous withdrawal rate = 40 gpm
8 hours operation, 5 days a week
Total estimated withdrawal during collection of the bulk sample = 1.15 MG
Estimated operation days = @ 45

### Environmental Evaluations for Permitting the Collection of the Bulk Sample



#### Required Permits & Environmental Reviews

- Environmental Assement Worksheet (DNR is RGU)
- Water Appropriations Permit DNR
- > Operation & Reclamation Plan DNR
- State Disposal System Permit (for stormwater rapid infiltration basin) – MPCA
- Conditional Use Permit City of Emily
- Underground Injection Control Permit EPA

### Aquifer Testing and Modeling



Photo by Tonia O'Brien, Barr Engineering Co.

## Drilling 18-inch Borehole





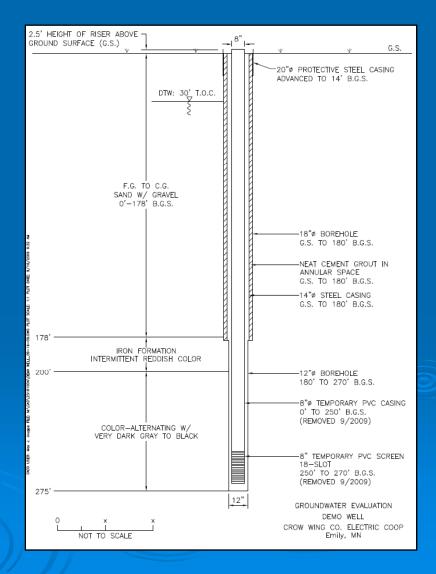
14-inch Steel Casing

Photo by Ellen Considine, Barr Engineering Company



#### Photo by Ellen Considine, Barr Engineering Company

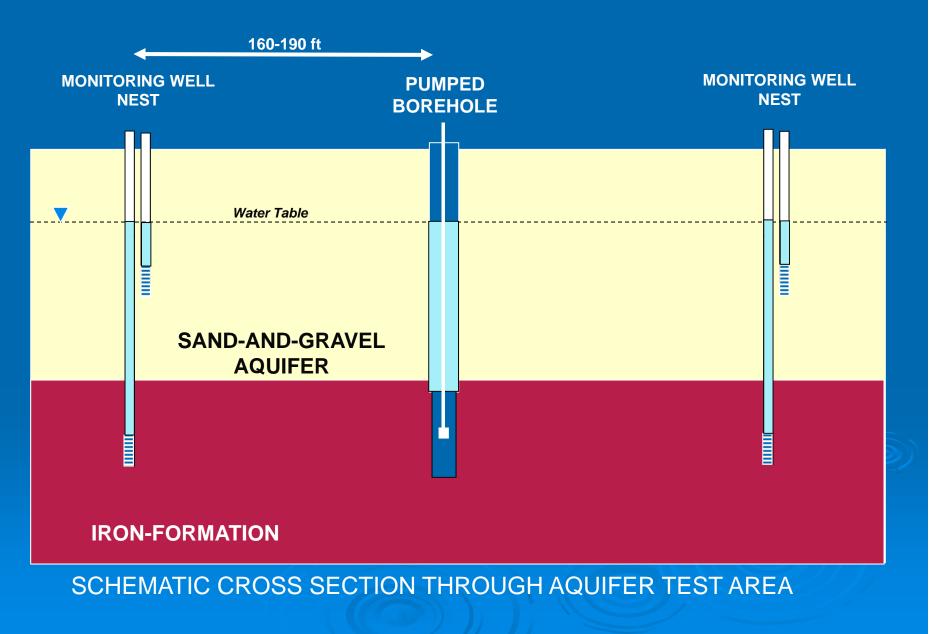
#### 8-inch Temporary Casing Installed in Borehole

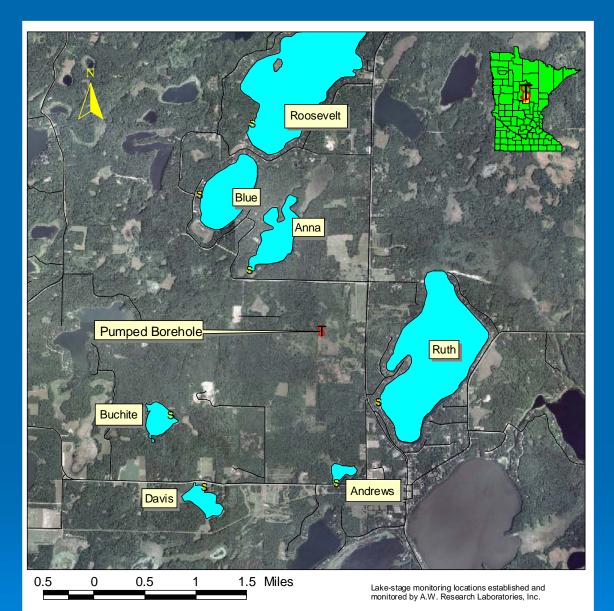




#### Location of Monitoring Wells

#### **Non-Pumping Condition**





Lake Stage and Pan Evaporation were also monitored before, during, and after test



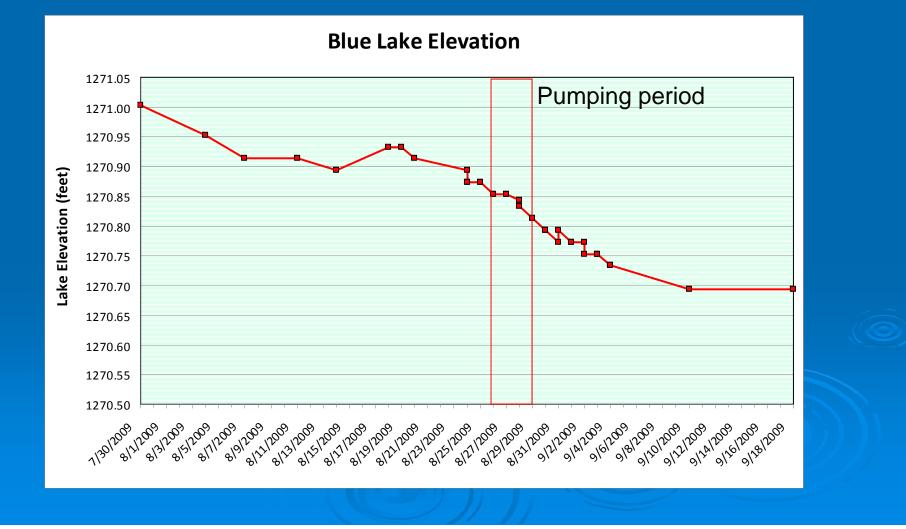
Photo by A.W. Research, Inc.

S Monitoring Location

# Water chemistry in both units are very similar

 $\succ$  Dissolved Iron – 0.02 to 23 mg/L – typically higher in sand-and-gravel aquifer  $\succ$  Dissolved Mn – 0.086 to 1.4 mg/L typically higher in sand-and-gravel aquifer > Major ions @ similar concentrations in both units > Trace metals non-detect to a few ppb > pH @ 6.5 to 7 > Eh @ -140 mv in both units

### Example of Lake Stage Trends

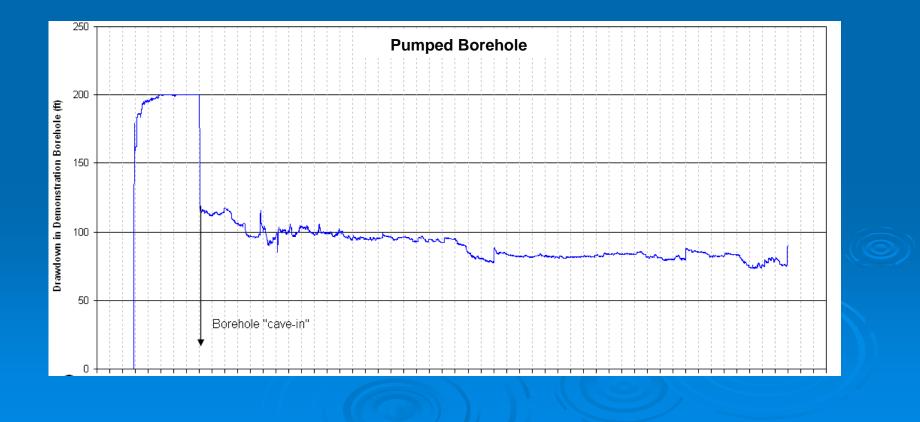


## Aquifer Test

> 162.7 hours of continuous pumping @ +/-200 gpm

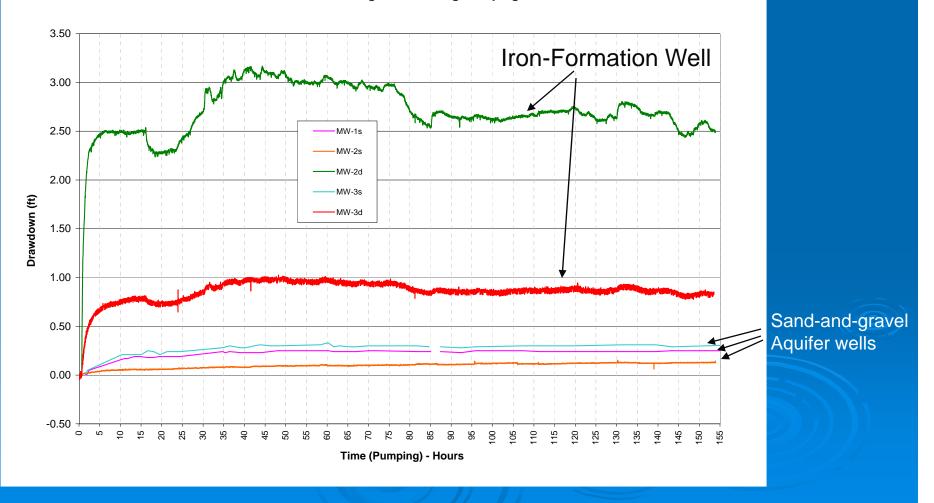
- > 1.95 million gallons of water
- That's about 0.8 million gallons more than will likely be pumped by the entire Bulk Sample Collection Project

## Borehole Cave-In Resulted in Huge Improvement in Well Efficiency



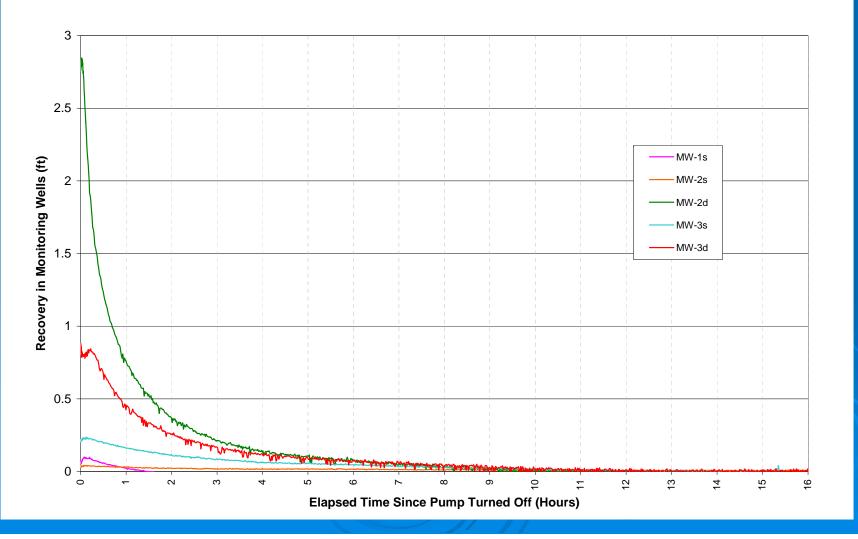
### Maximum Drawdown in the Sand-and-Gravel Aquifer Monitoring Wells was less than 0.5 feet

Water Levels in Monitoring Wells During Pumping Phase

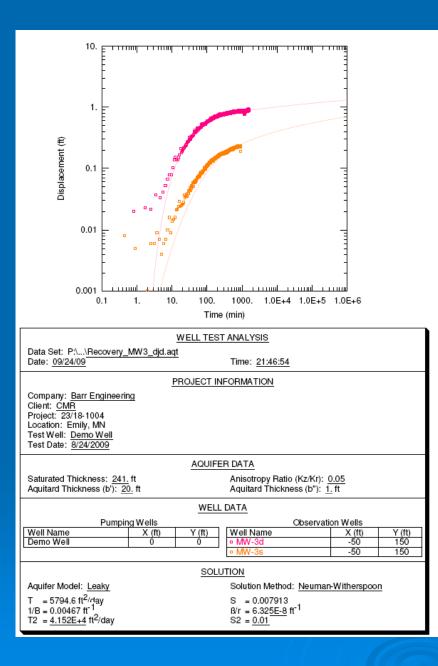


### Water-Level Recovery was Rapid

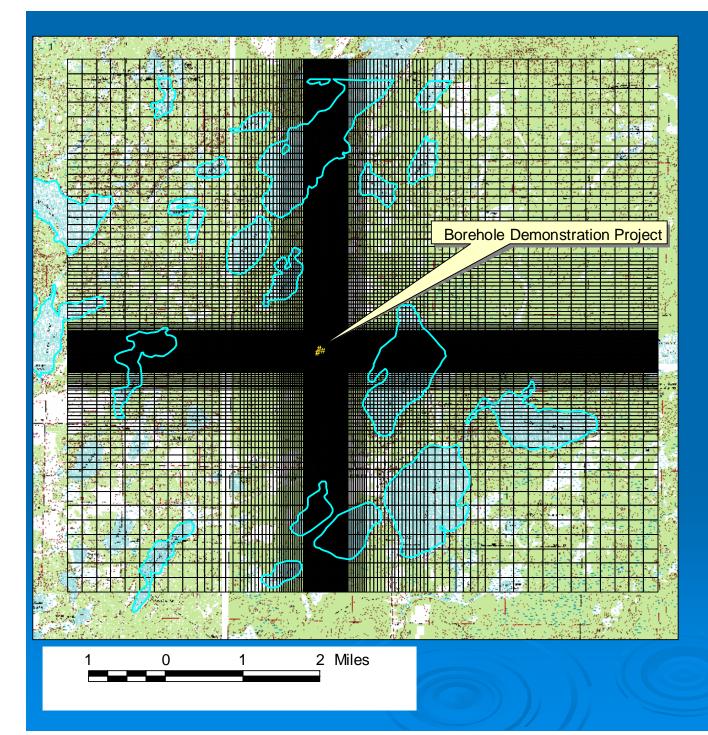
Water Level Recovery in Monitoring Wells



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Well pairs were analyzed for aquifer parameters using conventional analytic methods

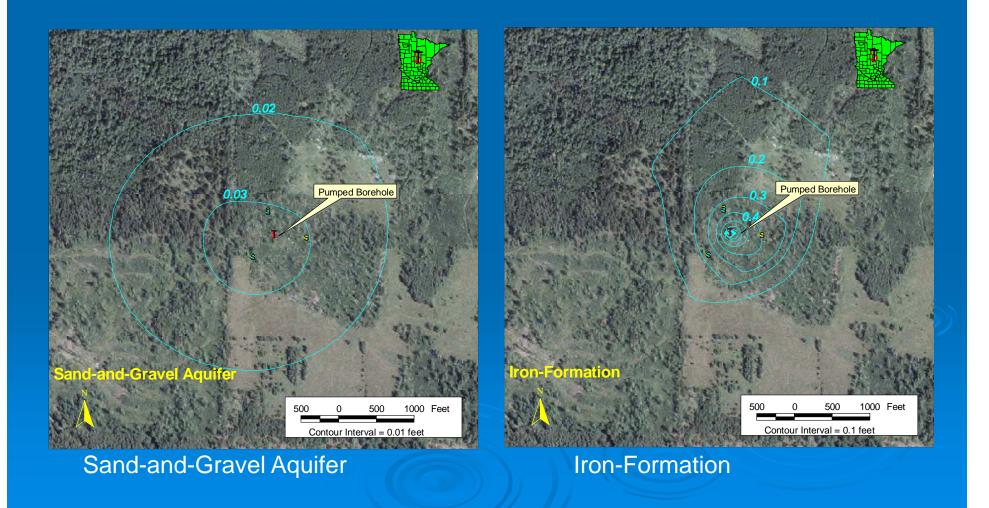


A 2-Layer MODFLOW model was developed and calibrated to the aquifer test data

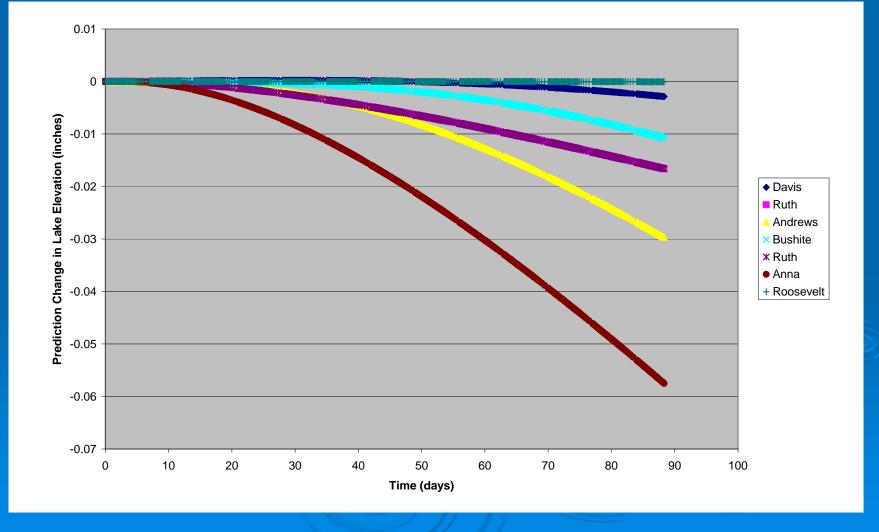
#### Model Calibration Designed to "Honor" Analytic Solution Results

Model Parame ter	Parameter Type	Model Calibration Value	Neuman-Witherspoon Value
kz6	vertical K of iron-formation @ MW-3D	0.0058 ft/day	0.00590 ft/day (based on 1/B)
kz7	vertical K of iron-formation @ MW-2D	12.4 ft/day	16.7 ft/day (based on 1/B)
kz8	vertical K of iron-formation @ MW-1S	2.1 ft/day	1.60 ft/day (based on 1/B)
sy1	specific yield of sand-and- gravel aquifer	0.08 ft/day	0.01 - 0.04
s2	storativity of bedrock units	0.0049	0.0001 - 0.0079
kx6	horizontal K of iron- formation @ MW-3D	24.4 ft/day	24 ft/day
kx7	horizontal K of iron- formation @ MW-2D	18.3 ft/day	16 ft/day
kx8	horizontal K of iron- formation @ MW-1S	24.8 ft/day	25 ft/day
kx5	horizontal K of sand-and- gravel aquifer in vicinity of Bulk Sample Collection Project	293.6 ft/day	267 to 333 ft/day

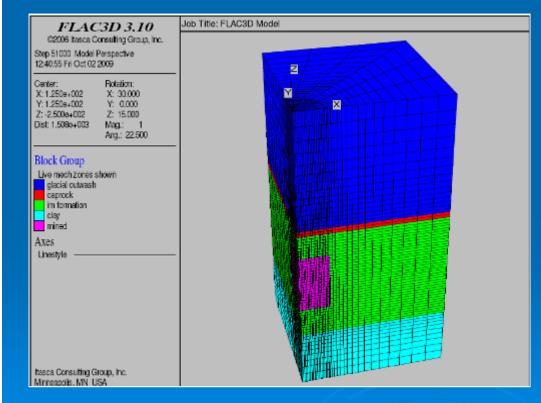
## Predicted Maximum Drawdown During the Bulk Sample Collection



### Model's Prediction of Changes in Lake Stage Elevation due to Pumping

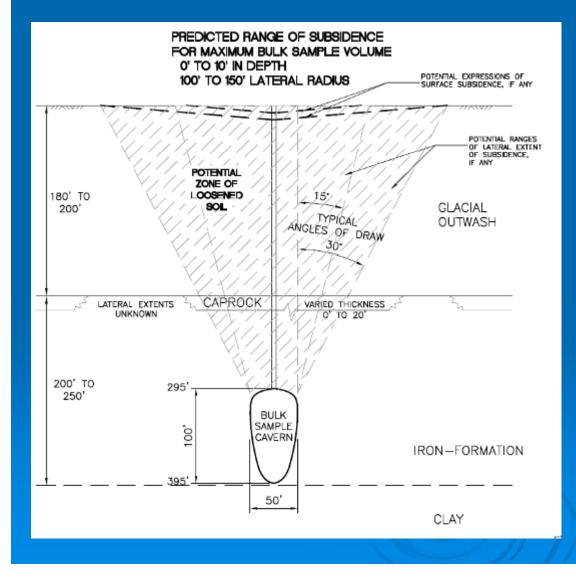


## Subsidence



 Modeled using program FLAC
 Model used geotechnical parameters from core tests

# Subsidence will be monitored during bulk sample collection



Maximum subsidence predicted to be <10 feet</p>

- Subsidence radius estimated to be <150 ft</li>
- Extensiometers will be installed to monitor subsidence

### What's Next

- Bulk sampling is planned to begin in August
- Continuous monitoring will occur during the sample collection for hydrologic and geotechnical conditions – useful for an EIS
- Full-scale mining will depend on what is learned during the bulk sample collection project – an EIS will be required for this
- Full-scale mining will likely be 1 to 4 simultaneous boreholes in operation at any one time

## Acknowledgments

### Ray Wuolo– BARR Engineering

### James Agre - Crow Wing Power

## QUESTIONS ?????



### please visit our web site:

### mndnr.gov/waters

