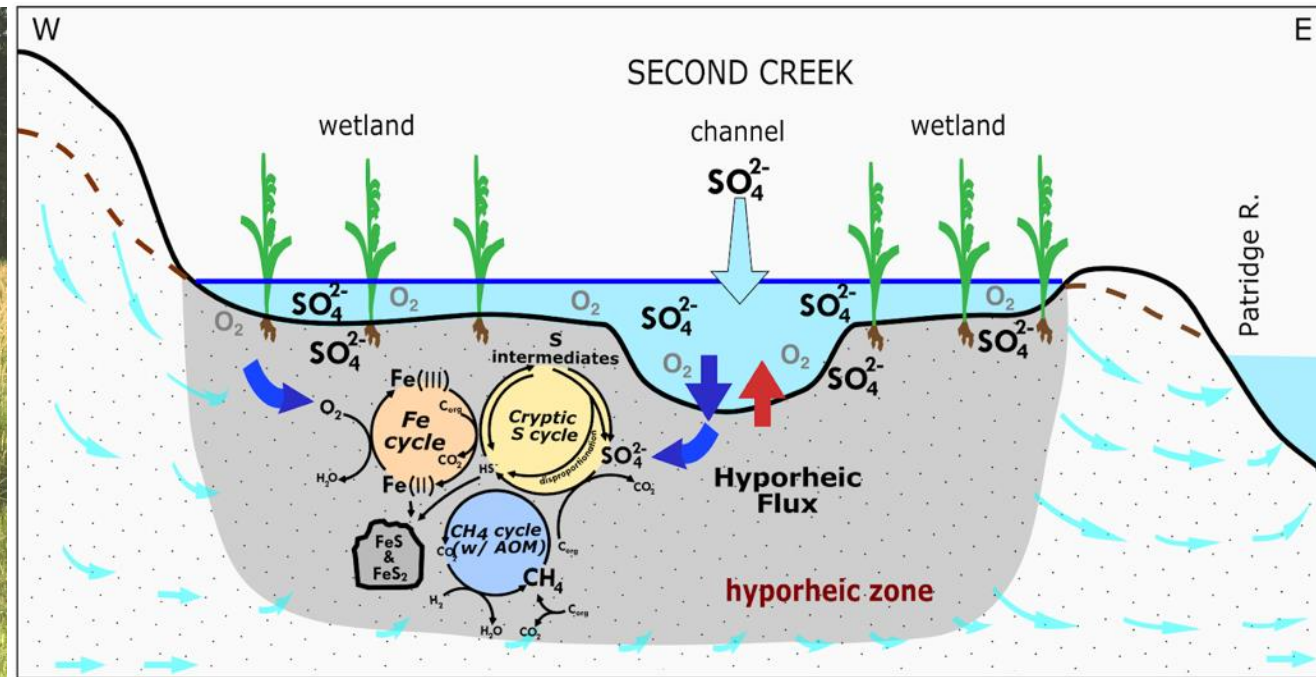


# Uncovering hidden sulfur biogeochemical cycles in a sulfate-impacted riparian wetland and stream in northeastern Minnesota



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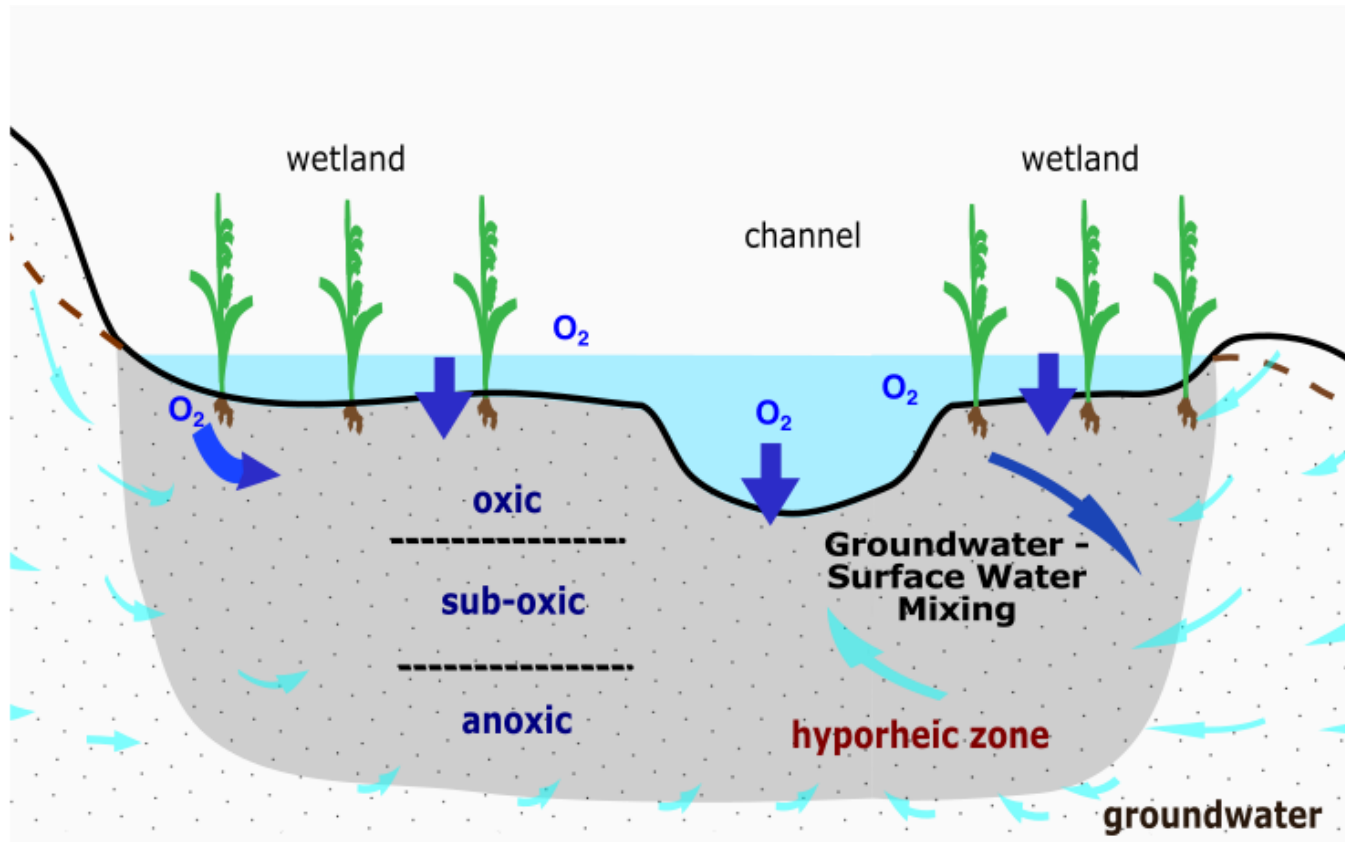
**Riley Schmitter**

**Patrick O'Hara**



# Hyporheic Zone Hydrobiogeochemistry

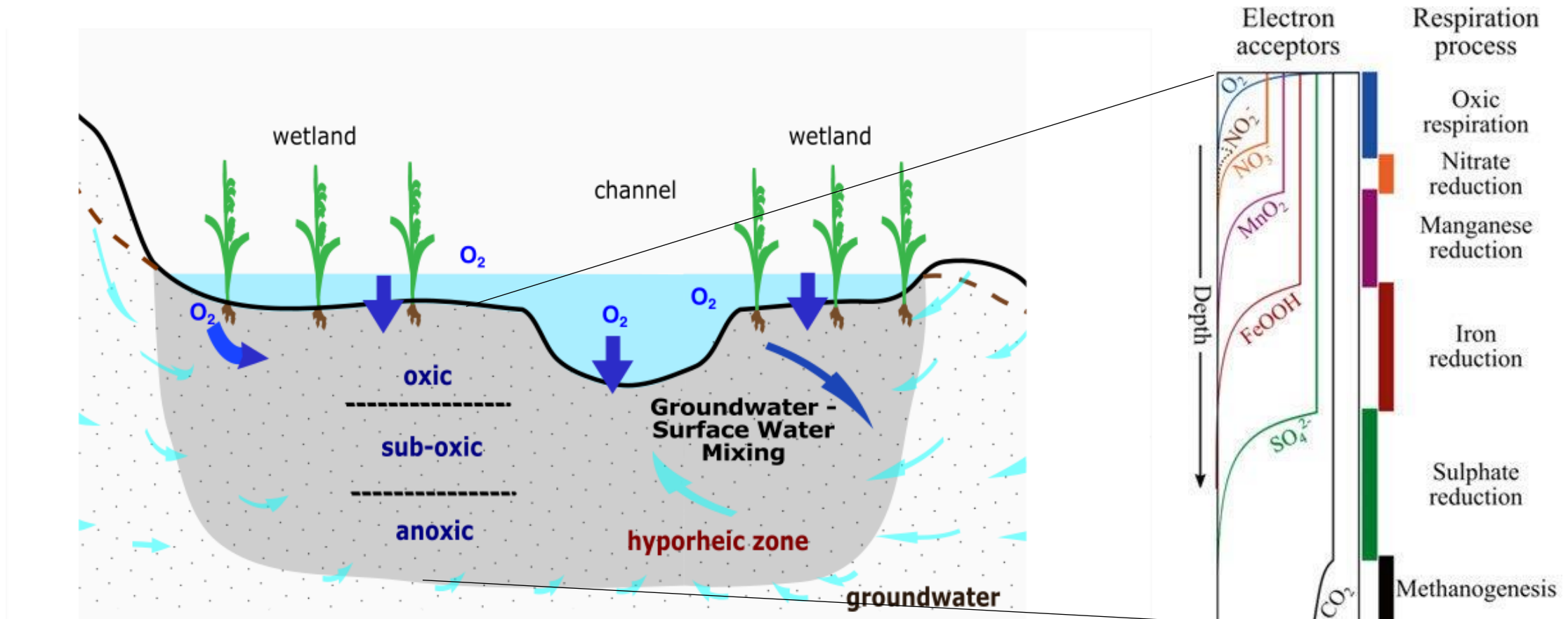
Biogeochemical cycling of nutrients/metals has important environmental impacts!



*Changes in flux magnitude and direction will impact redox gradients and biogeochemical cycling, and ultimately nutrient/contaminant fate and transport.*

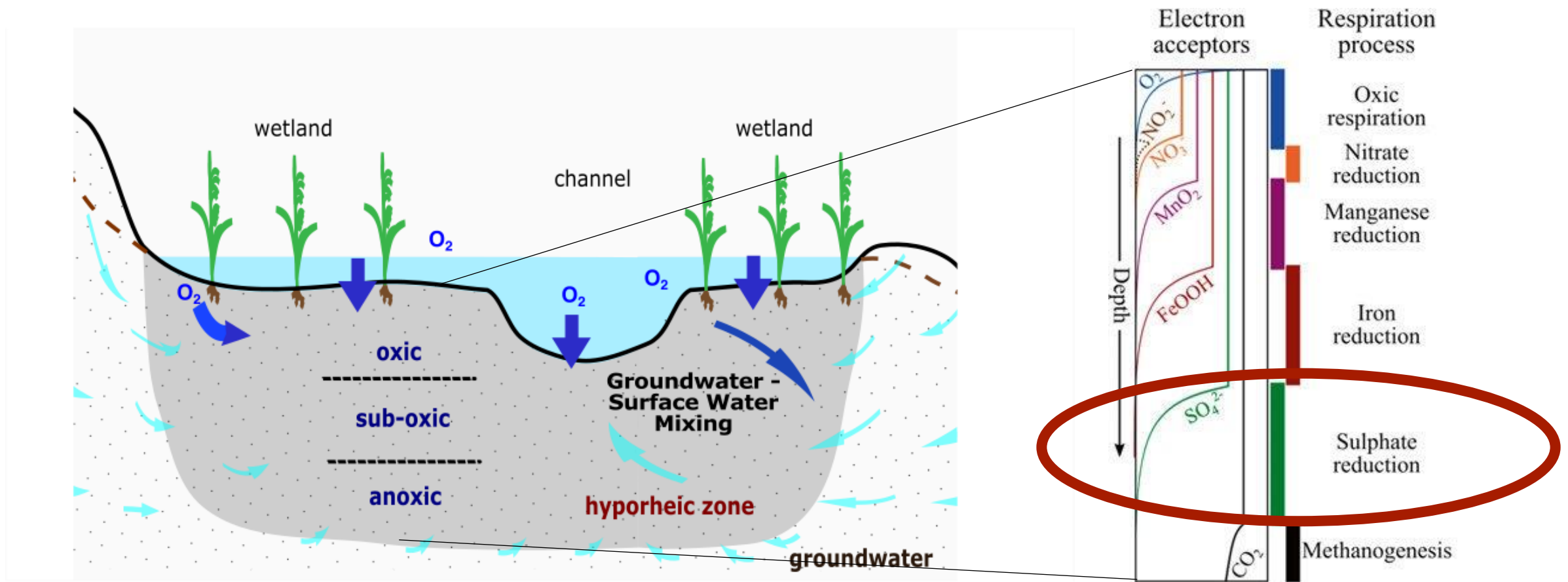
# Hyporheic Zone Hydrobiogeochemistry

Microbially-driven and abiotic geochemical oxidation-reduction “redox” reactions occur in a **gradient** across this zone. This gradient can be quite dynamic, influenced by water fluxes!



# Why is sulfur overlooked in freshwater environments?

Sulfate concentrations are low in most systems.

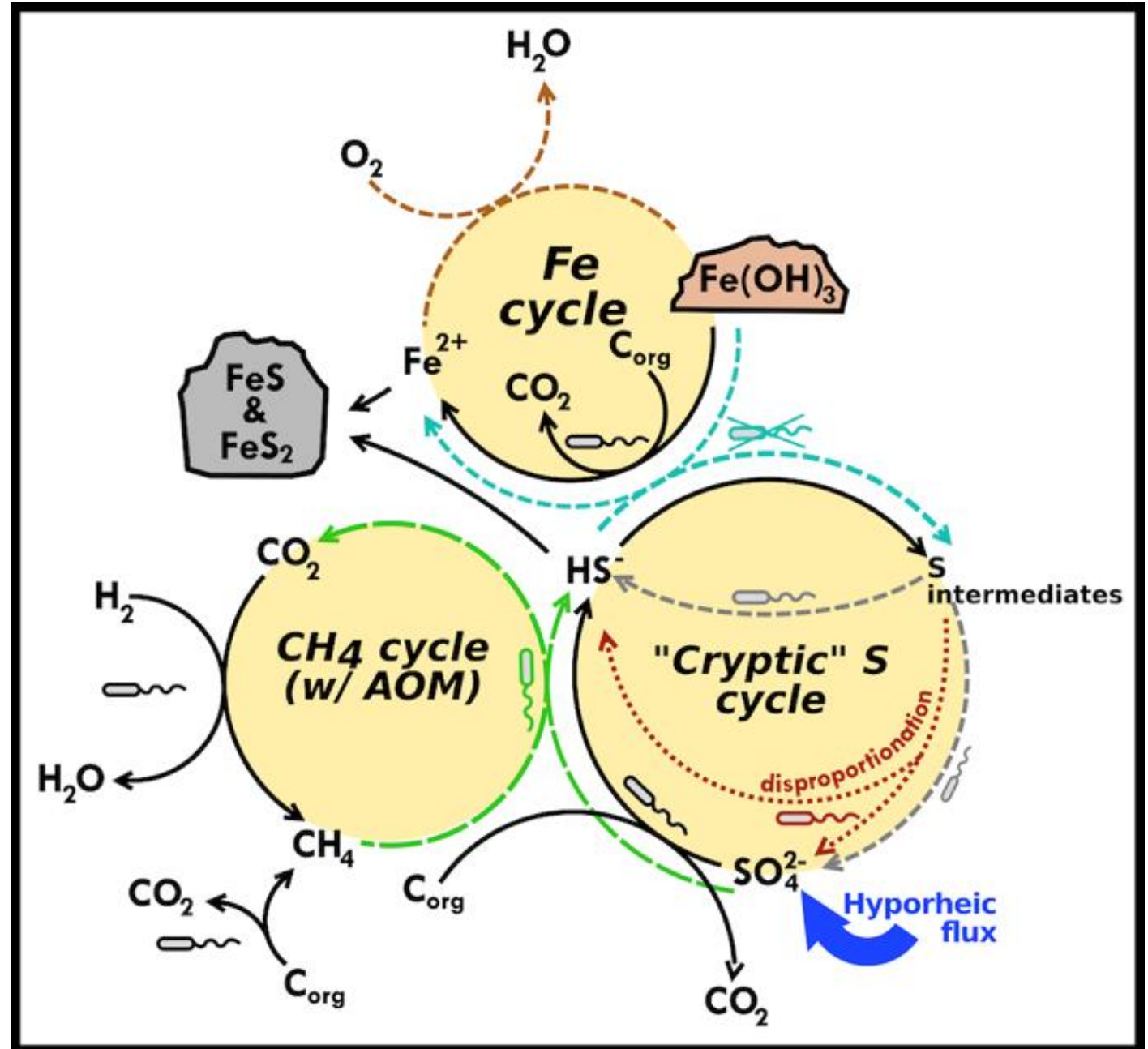


Steffen Weirs, 2020

*But S inputs increasing due to agriculture (Hinckley et al., 2020) and industry.*

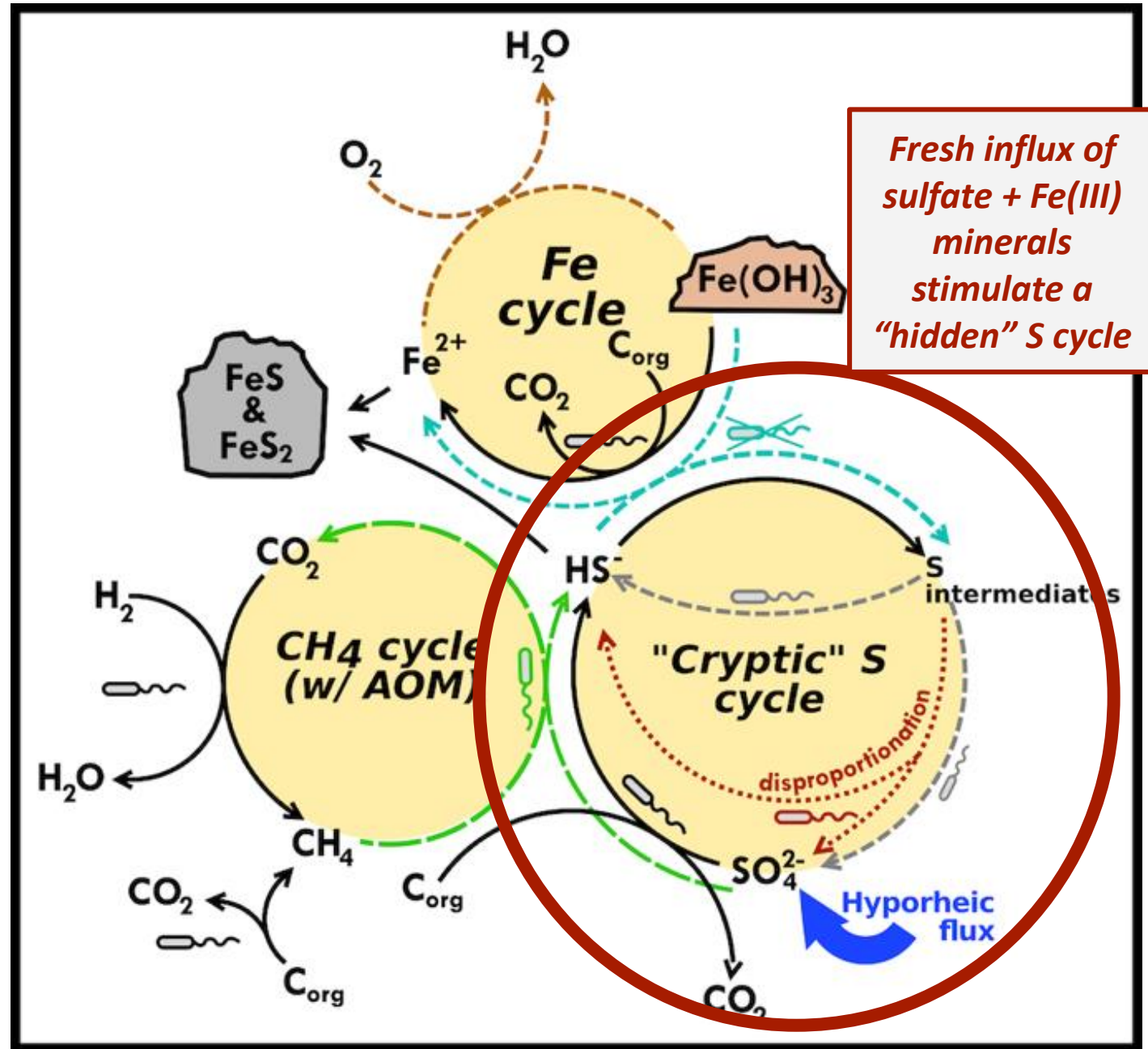
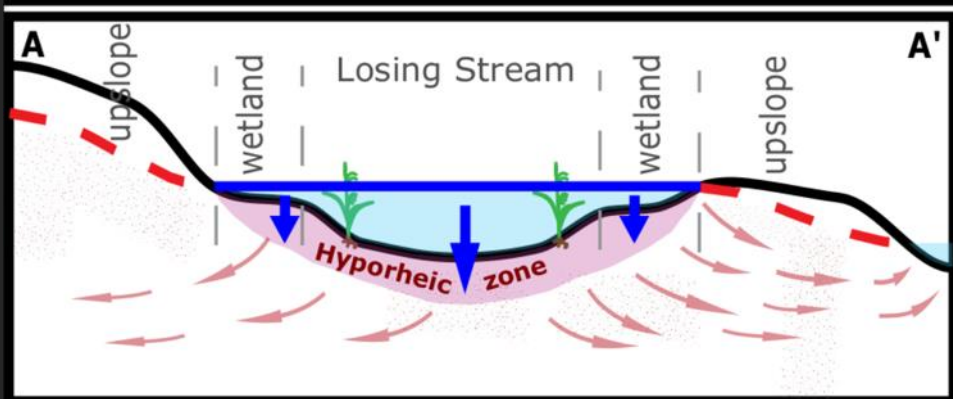
# Why care about S cycling in freshwater wetlands and streams?

- S high/increasing in industrial (e.g., mining) and agricultural waters
- Coupled to Fe (nutrient) cycling and bioavailability
- Coupled to methane cycling and may impact emissions



# Linked Fe-S-C biogeochemical cycling in hyporheic zone stimulated by dynamic hydrologic fluxes

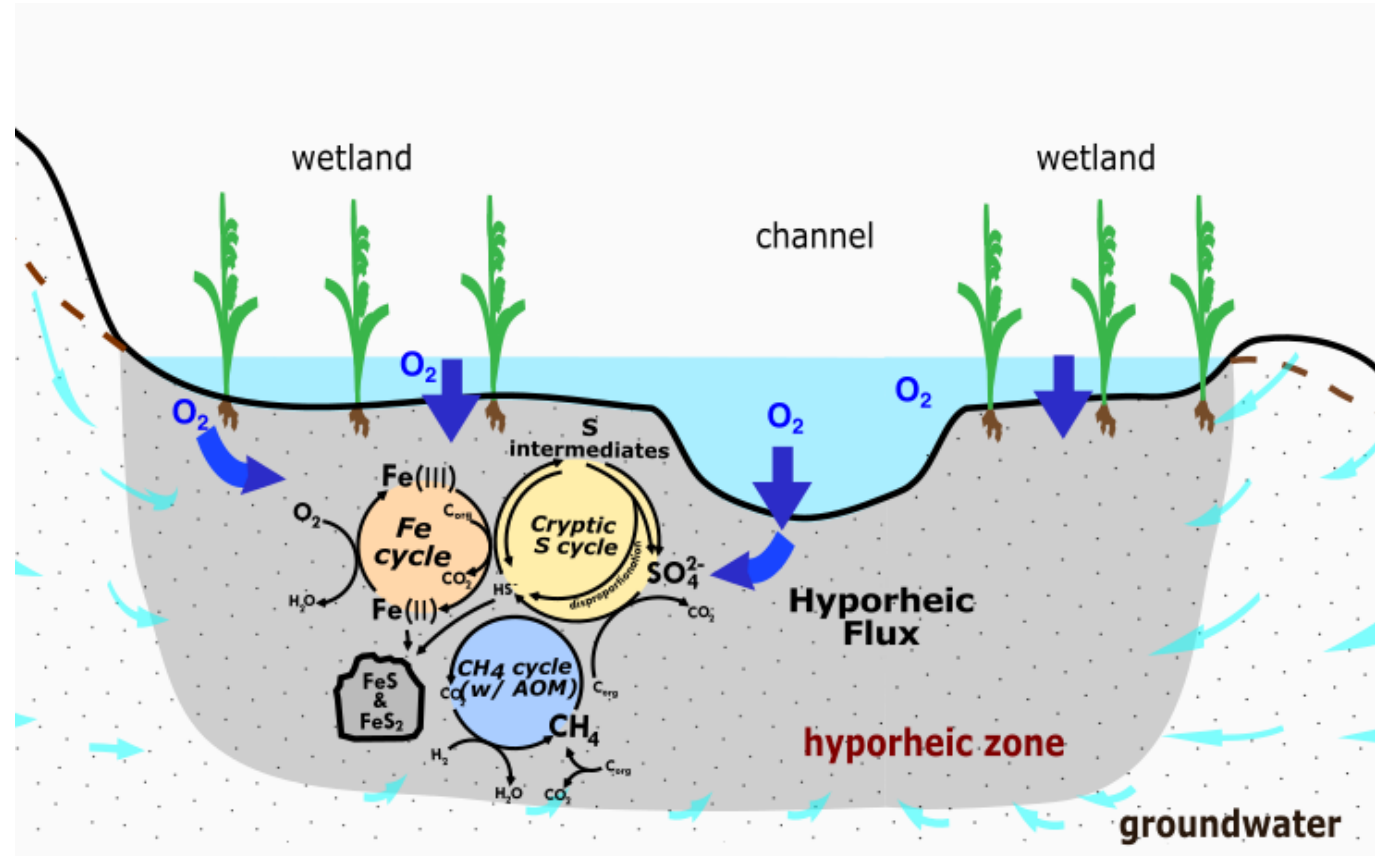
*Dynamic hydrologic fluxes can push oxygenated waters into subsurface, altering redox gradients*



# Objectives

Establish the connection between hyporheic flux and subsurface “cryptic” S cycle coupled to Fe and C redox cycling

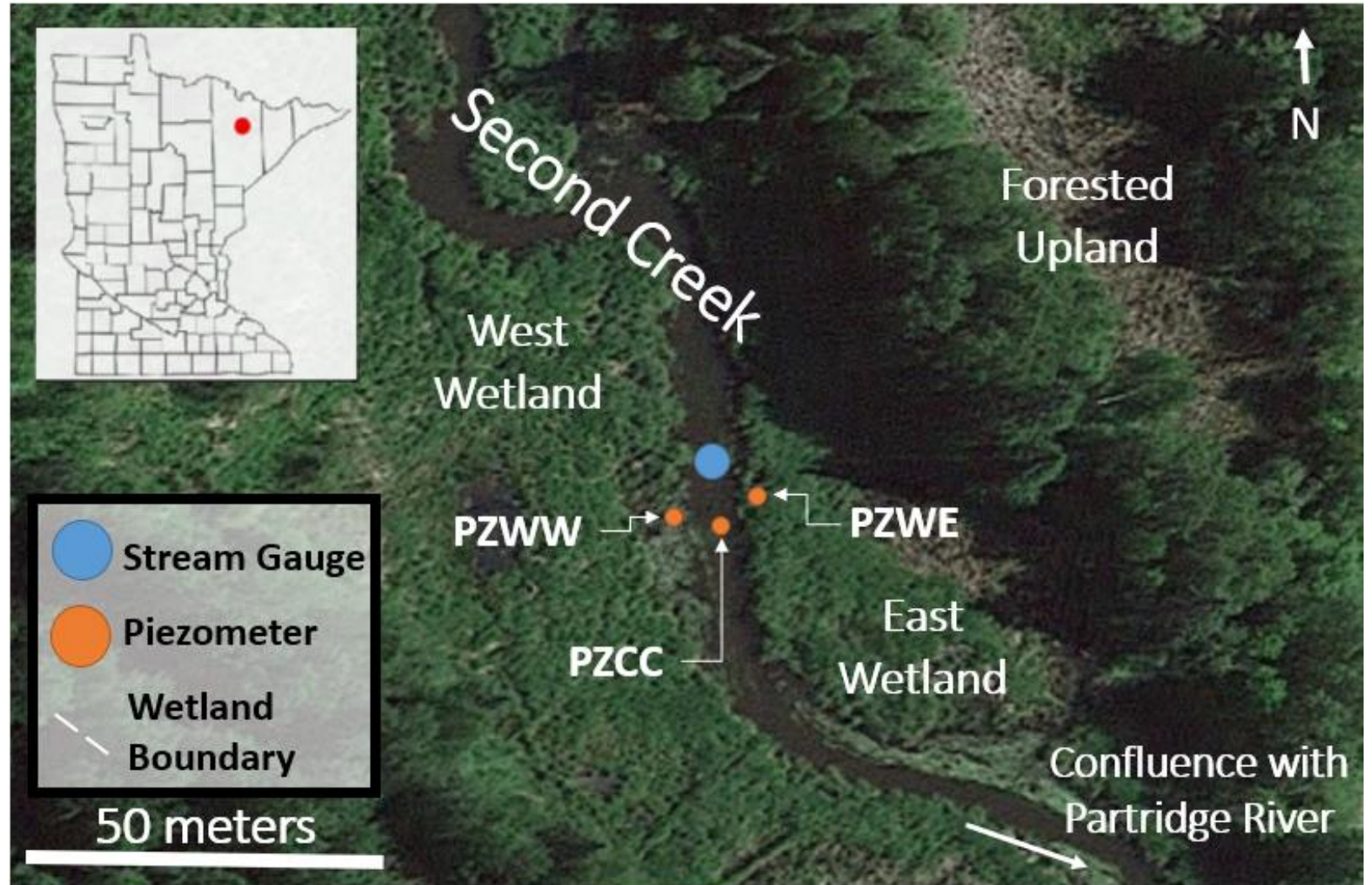
Determine products and impacts of high sulfate water due to cryptic S cycling



# Evidence for cryptic S cycling, influenced by dynamic hyporheic fluxes, in sulfate-impacted stream

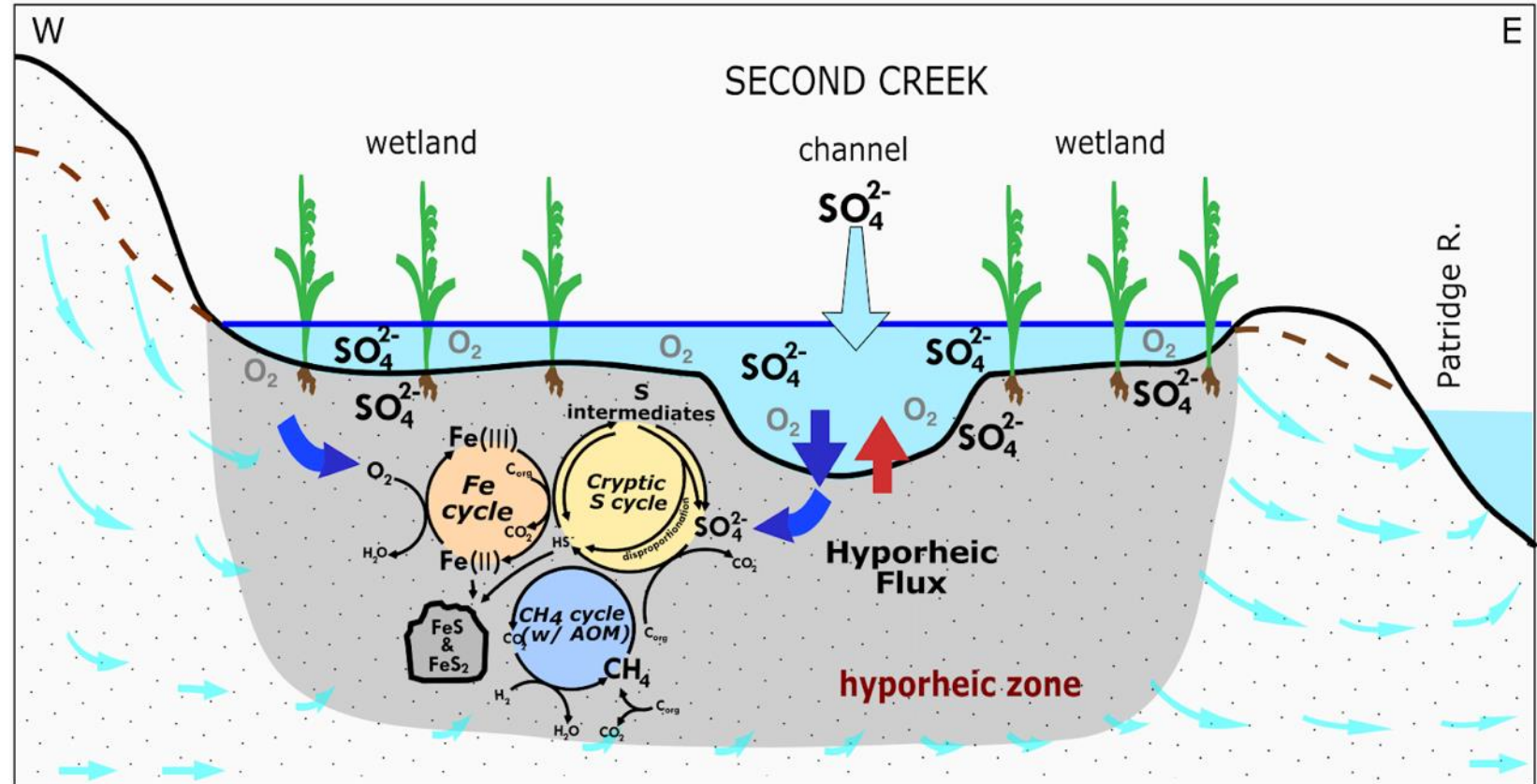
## Second Creek (Northern MN)

- Mixed boreal-deciduous ecotone
- Second Order stream just downstream of iron ore mining operations
- Fully inundated year round
- High sulfate (1-10 mM) in surface water
- Low nitrate in surface water
- High iron in subsurface
- pH ~7-8





# Second Creek riparian wetlands and stream channel



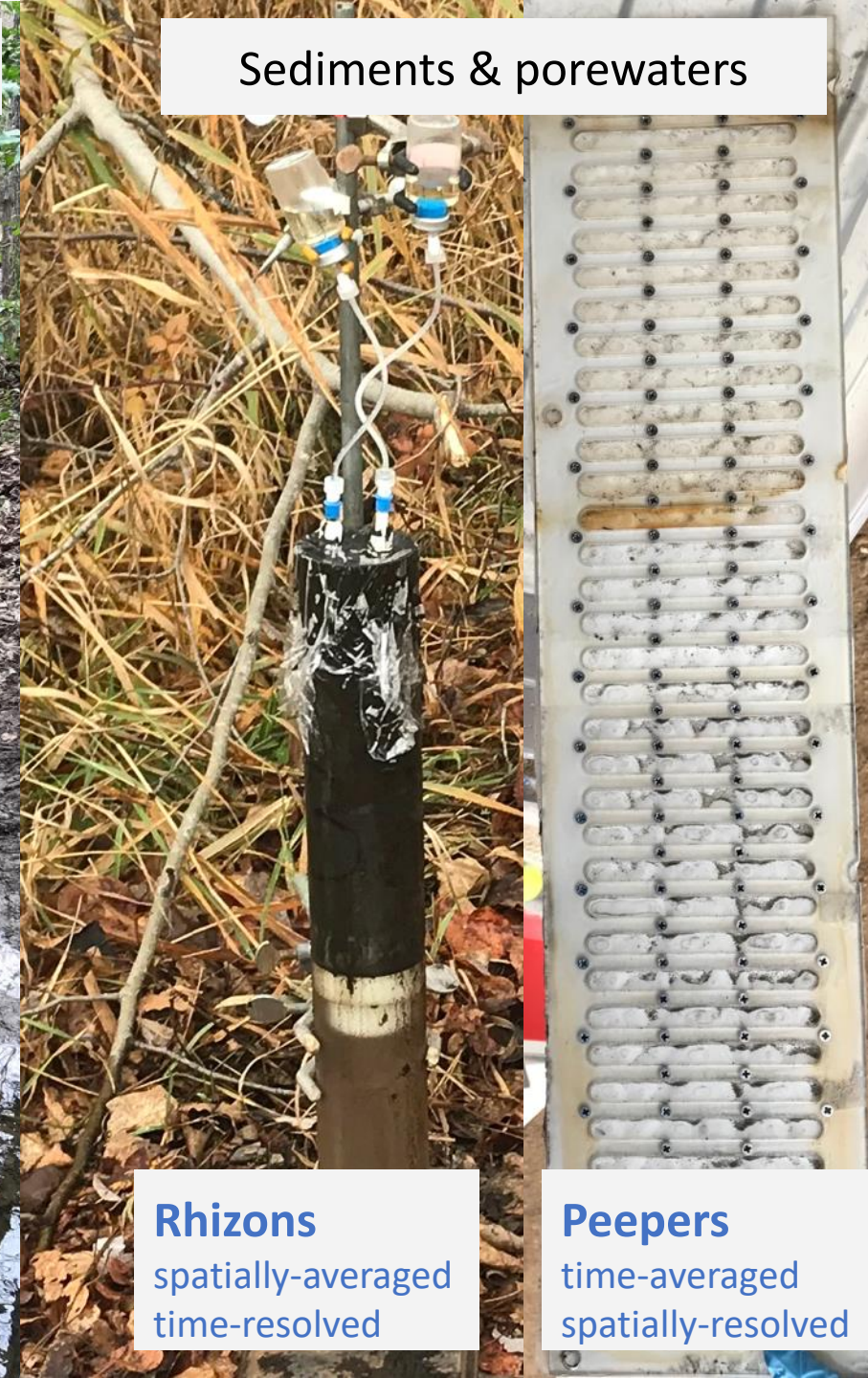
**Wetlands**  
*Grasses/Sedges*  
*Organic Carbon rich*  
*Fine particle size*

**Channel**  
*Sparse plants*  
*Moderate  $C_{org}$*   
*Moderate particle size*

Surface/groundwater monitoring



Sediments & porewaters



**Rhizons**  
spatially-averaged  
time-resolved

**Peepers**  
time-averaged  
spatially-resolved

# Objective

Establish the connection between hydraulic flux and occurrence of a subsurface “cryptic” S cycle coupled to Fe and C redox cycling

## **Multi-pronged approach**

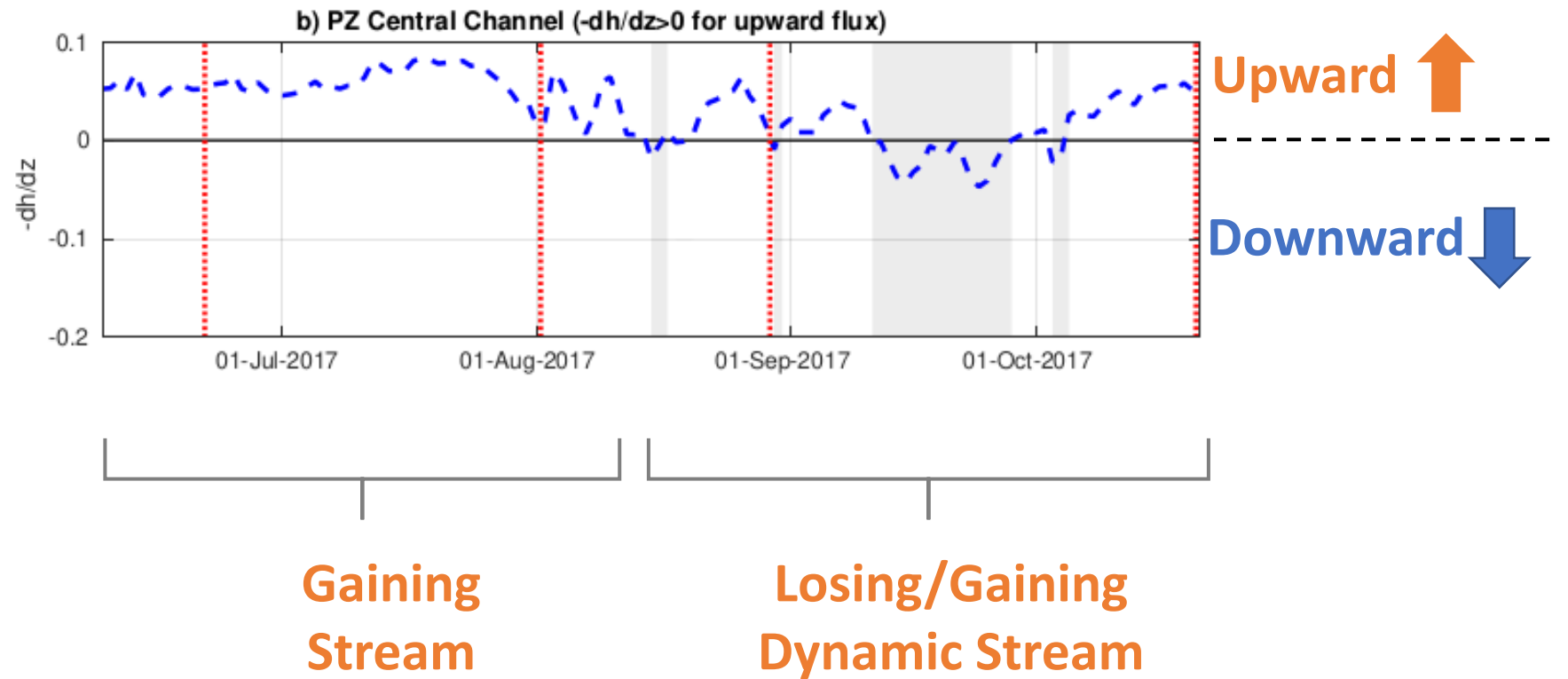
### Field observation:

- Hydrologic conditions
- Geochemistry
  - Surface water
  - Porewater
  - sediment

# Hyporheic exchange:

Are wetlands and stream gaining or losing?  
Change with time?

## Hydraulic Head Gradient (Central Stream Channel)



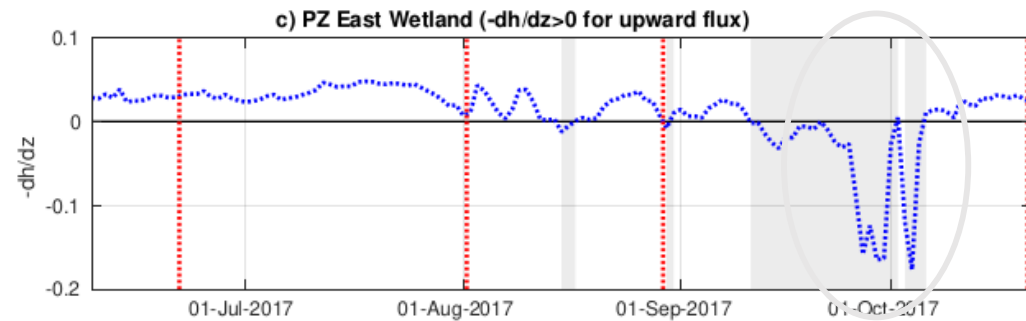
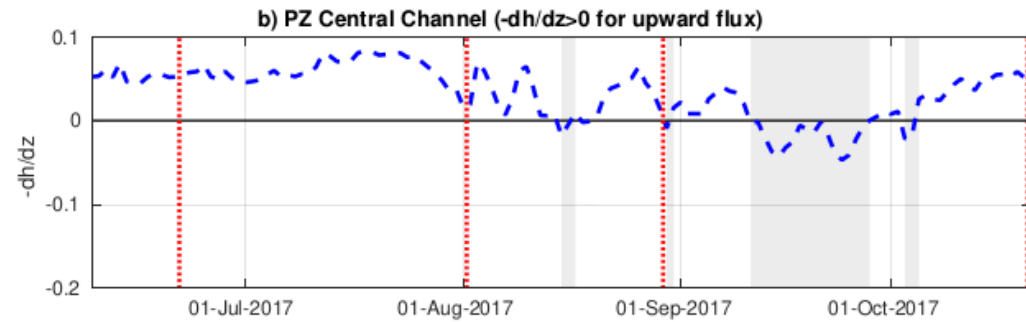
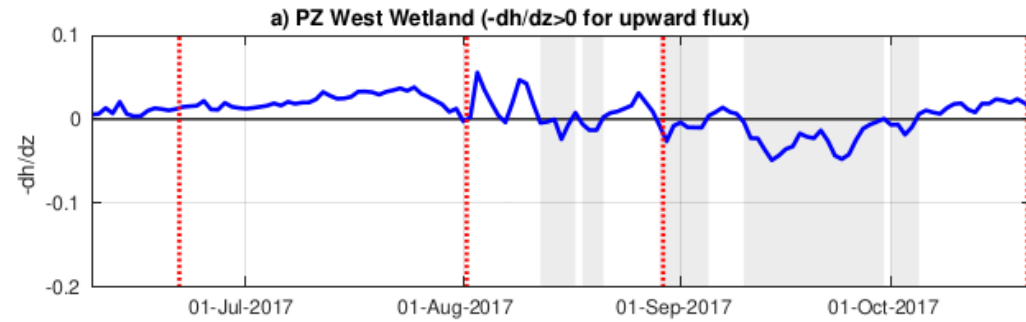
*Difficult to predict when to sample  
- telemetry monitoring beneficial*



Field sampling points

# Hyporheic exchange:

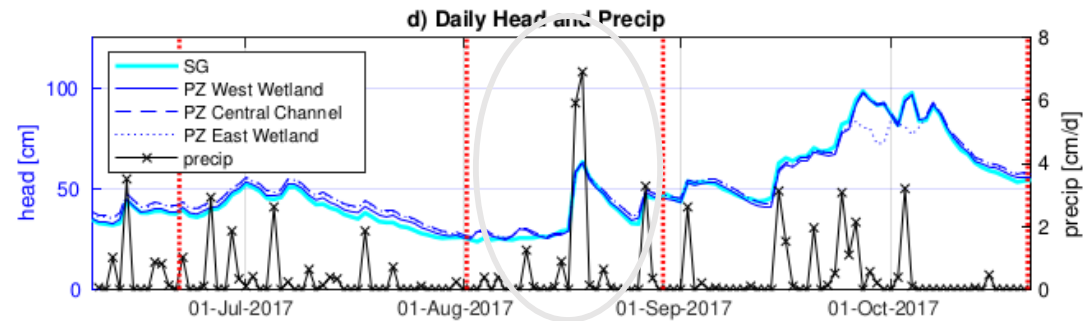
Are wetlands and stream gaining or losing?

## Hydraulic Head Gradient

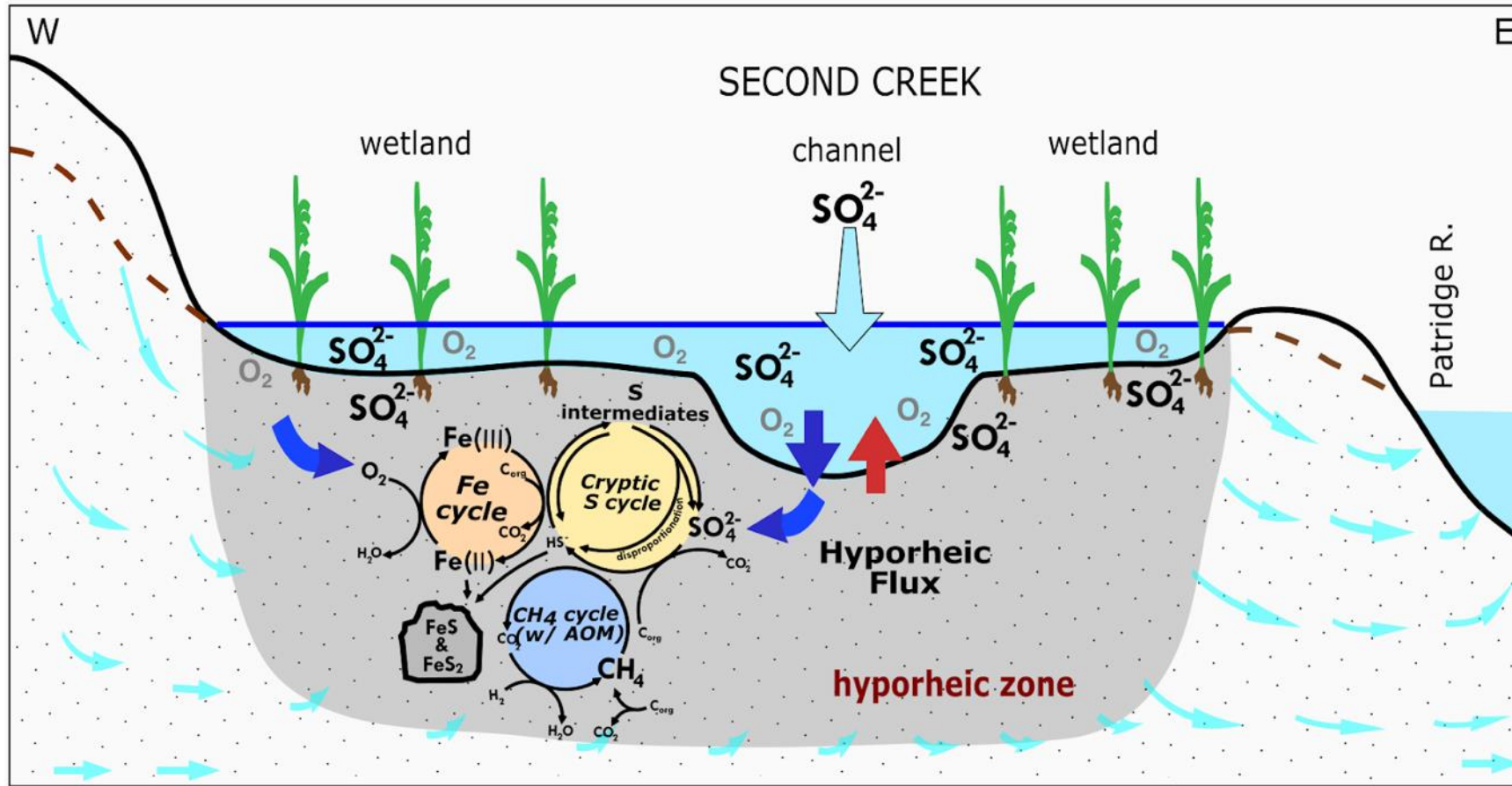


Gaining   
Losing 

*East Wetland saw greater magnitude downward flux*



*Dynamic flux response to high precipitation*



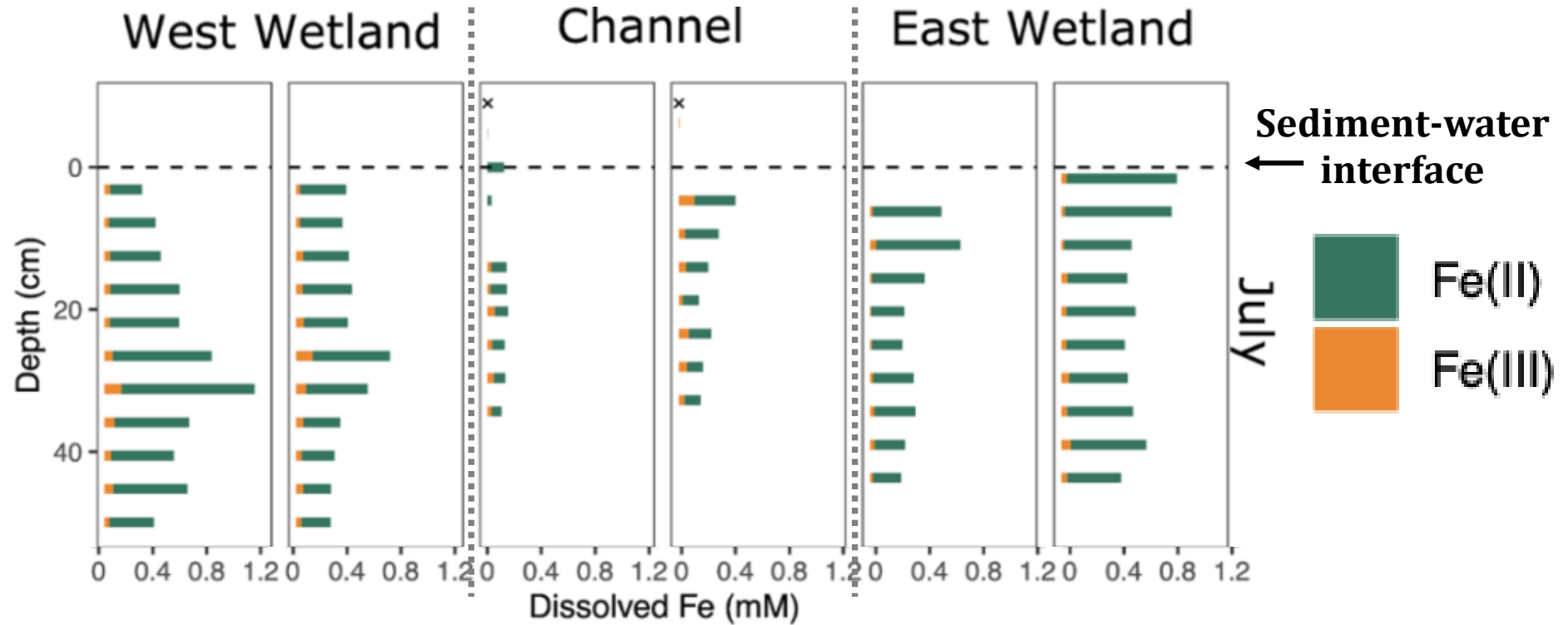
- *How does geochemistry respond to changes in hyporheic flux?*
- *Vary with location (wetlands vs. stream)?*
- *Evidence of Cryptic S cycle?*

# Porewater Aqueous Iron

(Fe<sup>2+</sup> + colloidal Fe<sup>3+</sup>)

High iron in  
porewater

Aqueous  
Fe(III)  
chelated by  
organics?



*Fe(II) dominant*  
*Fe(III) present at  
depth*

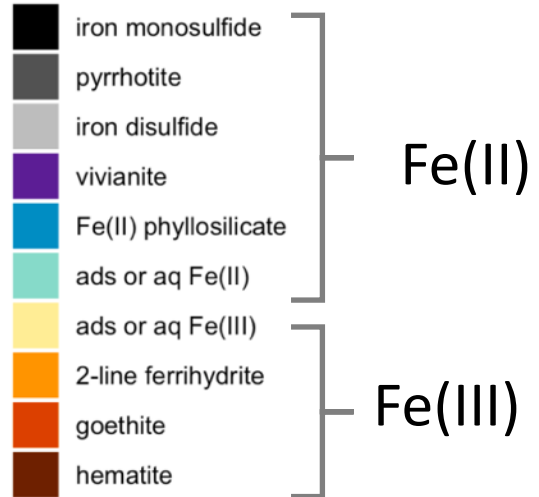
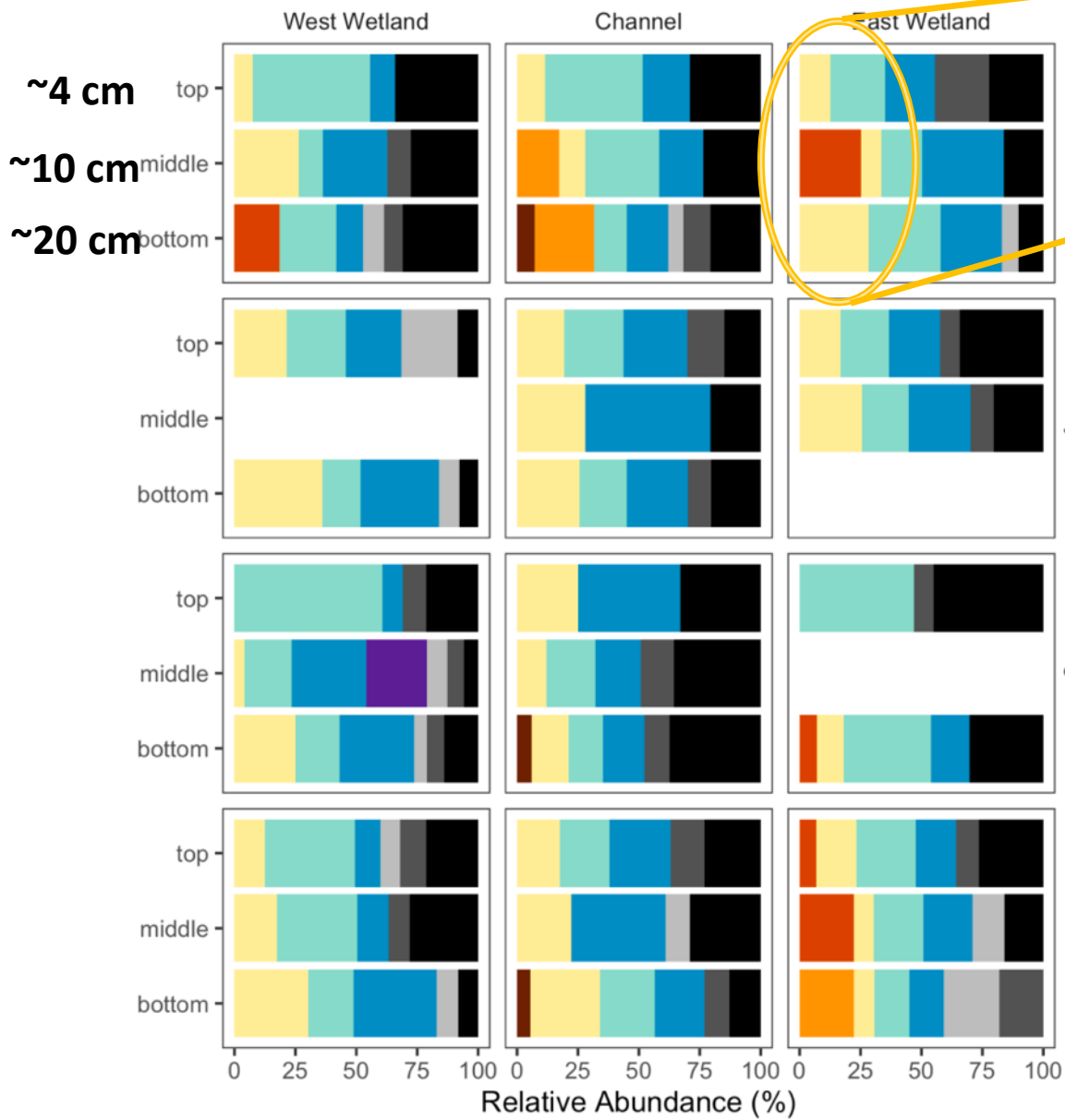
*Higher Fe<sub>total</sub>  
in wetlands*

*Trends  
consistent  
throughout  
summer/fall*

# Sediment Iron (Fe K-edge EXAFS)

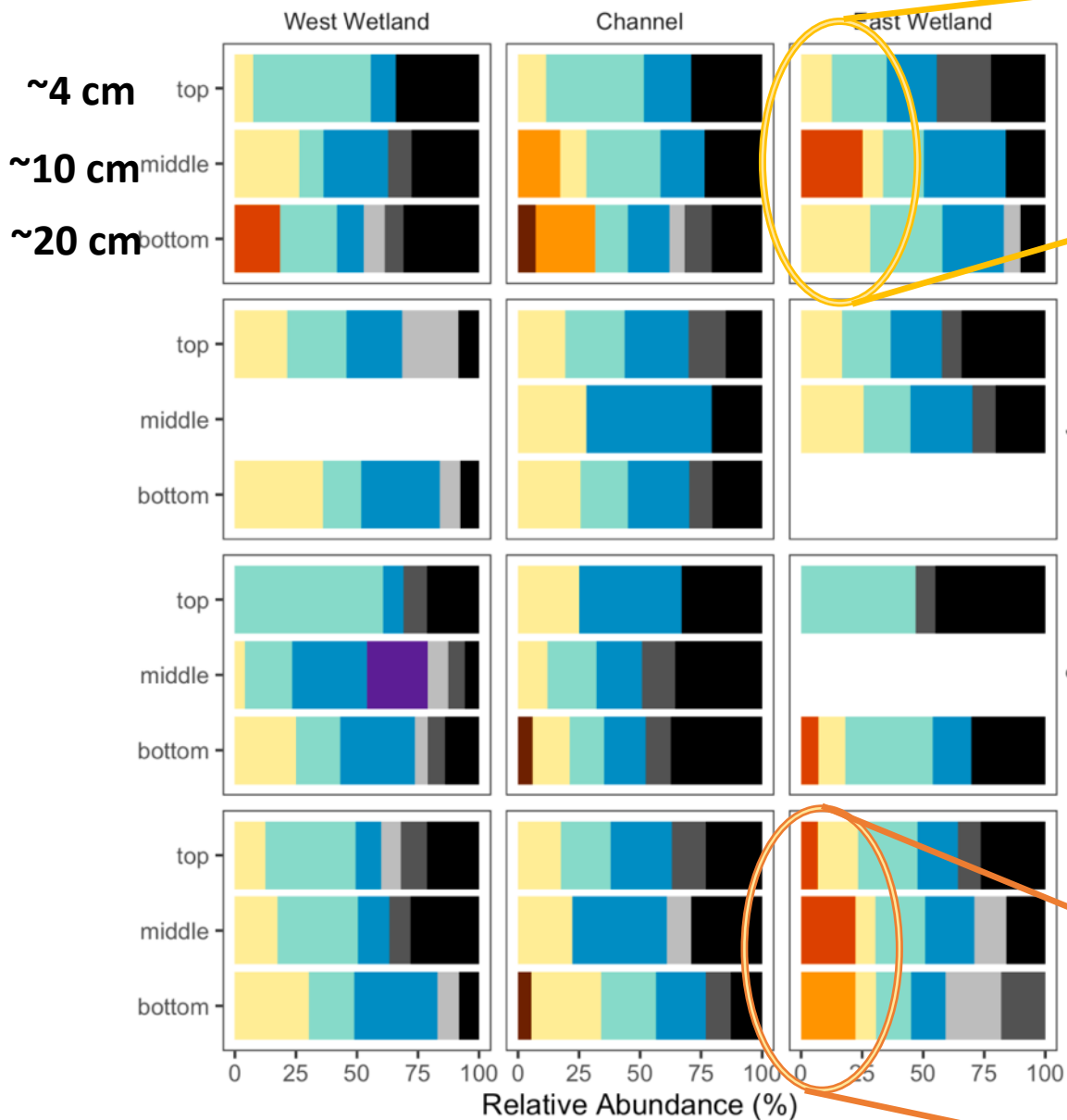
Fe(III) presence can help fuel a

*Fe(III)  
aqueous/adsorbed  
at depth*



# Sediment Iron (Fe K-edge EXAFS)

Fe(II) solid/aqueous phases dominate in anoxic sediments



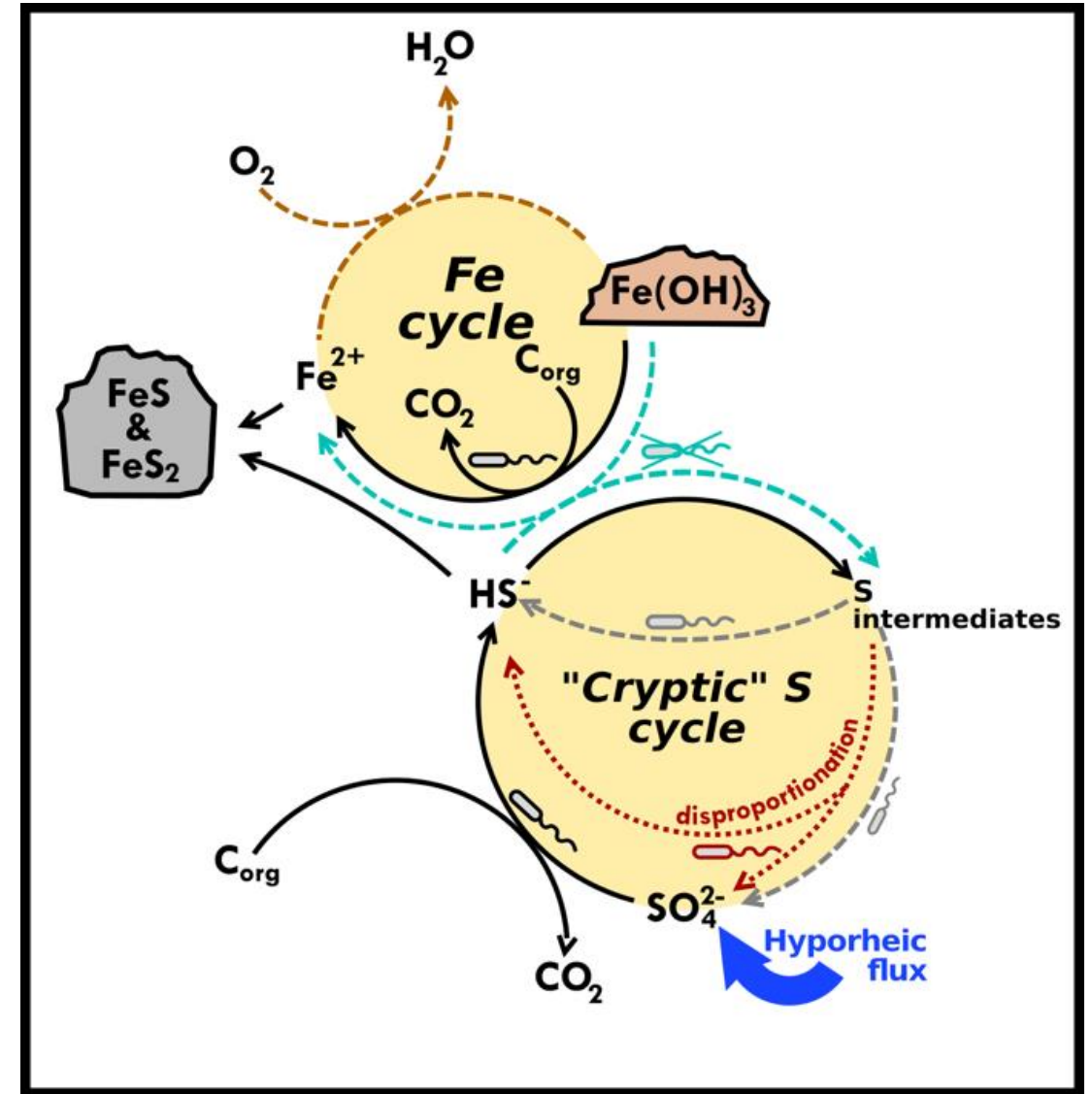
*Fe(III) aqueous/adsorbed at depth*

- iron monosulfide
  - pyrrhotite
  - iron disulfide
  - vivianite
  - Fe(II) phyllosilicate
  - ads or aq Fe(II)
  - ads or aq Fe(III)
  - 2-line ferrihydrite
  - goethite
  - hematite
- Fe(II) (includes: iron monosulfide, pyrrhotite, iron disulfide, vivianite, Fe(II) phyllosilicate, ads or aq Fe(II))
- Fe(III) (includes: ads or aq Fe(III), 2-line ferrihydrite, goethite, hematite)

*Fe(III) oxides after sustained downward flux*

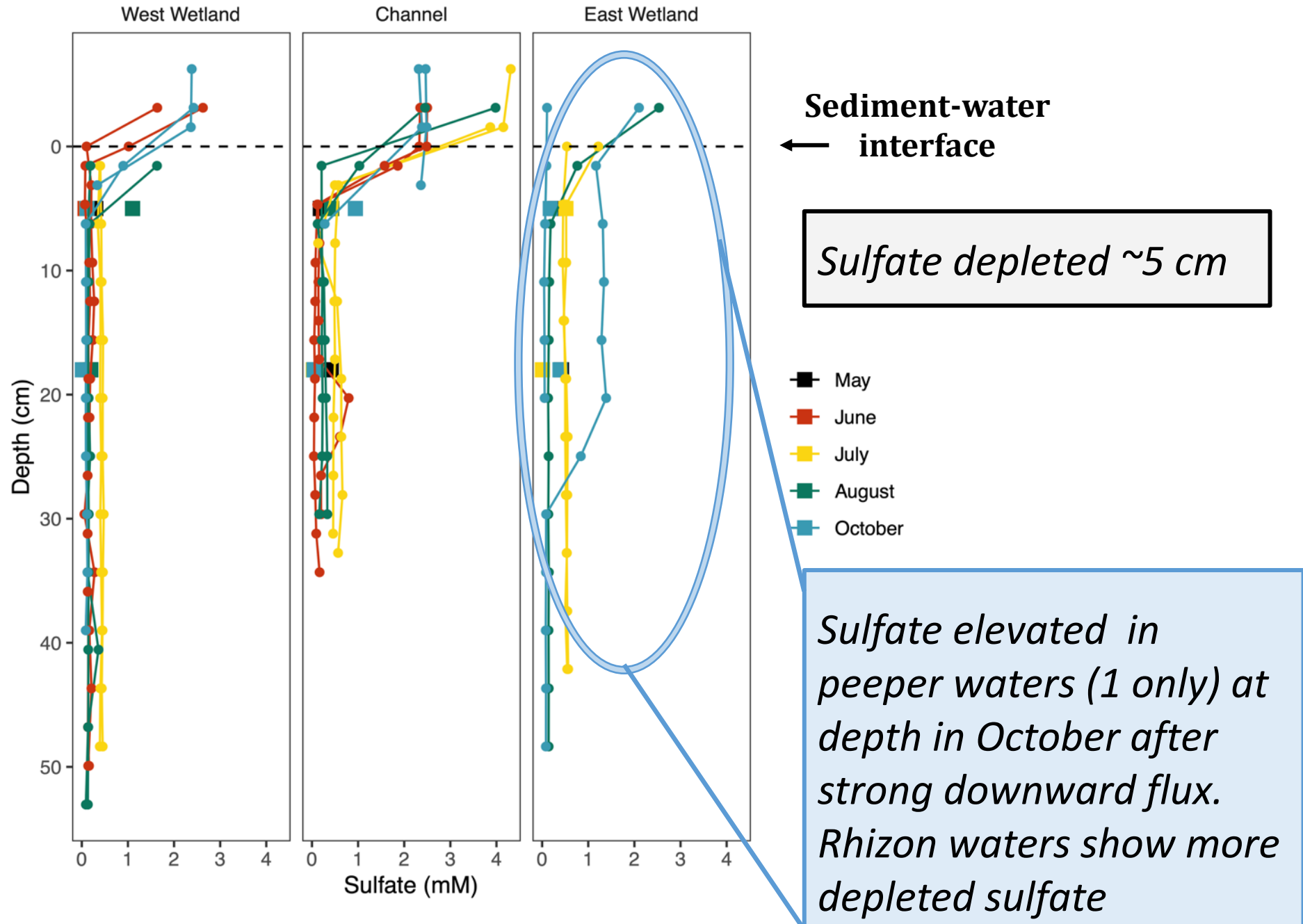


Can “cryptic” S cycle be sustained in anoxic sediments coupled to Fe(III) reduction?



# Aqueous sulfate

Sulfate reduced in upper sediments but levels remain >0 (~200  $\mu\text{M}$  average)

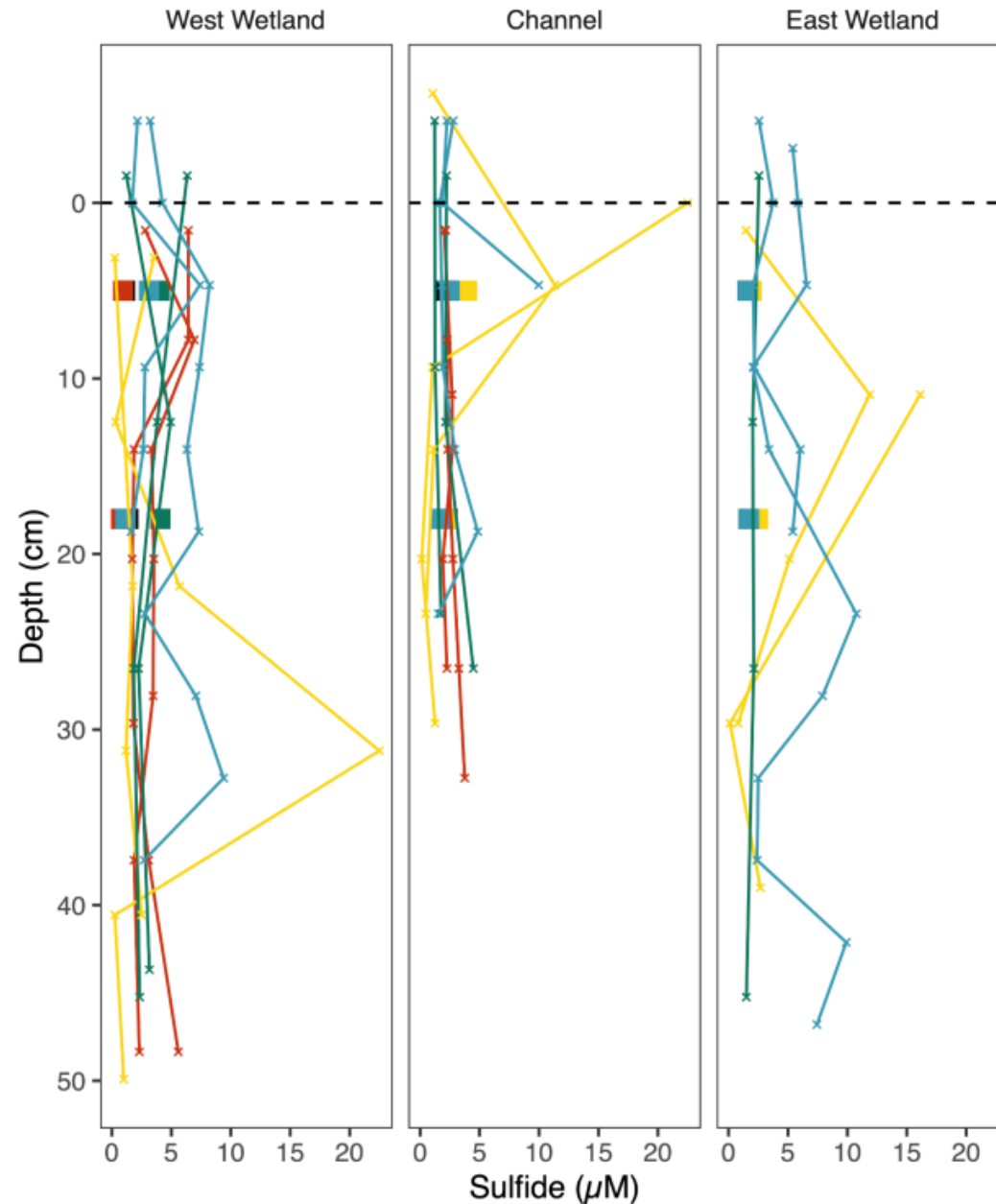


# Aqueous sulfide

(cline method)

Sulfate reduction to sulfide

NO major trends observed with depth or season

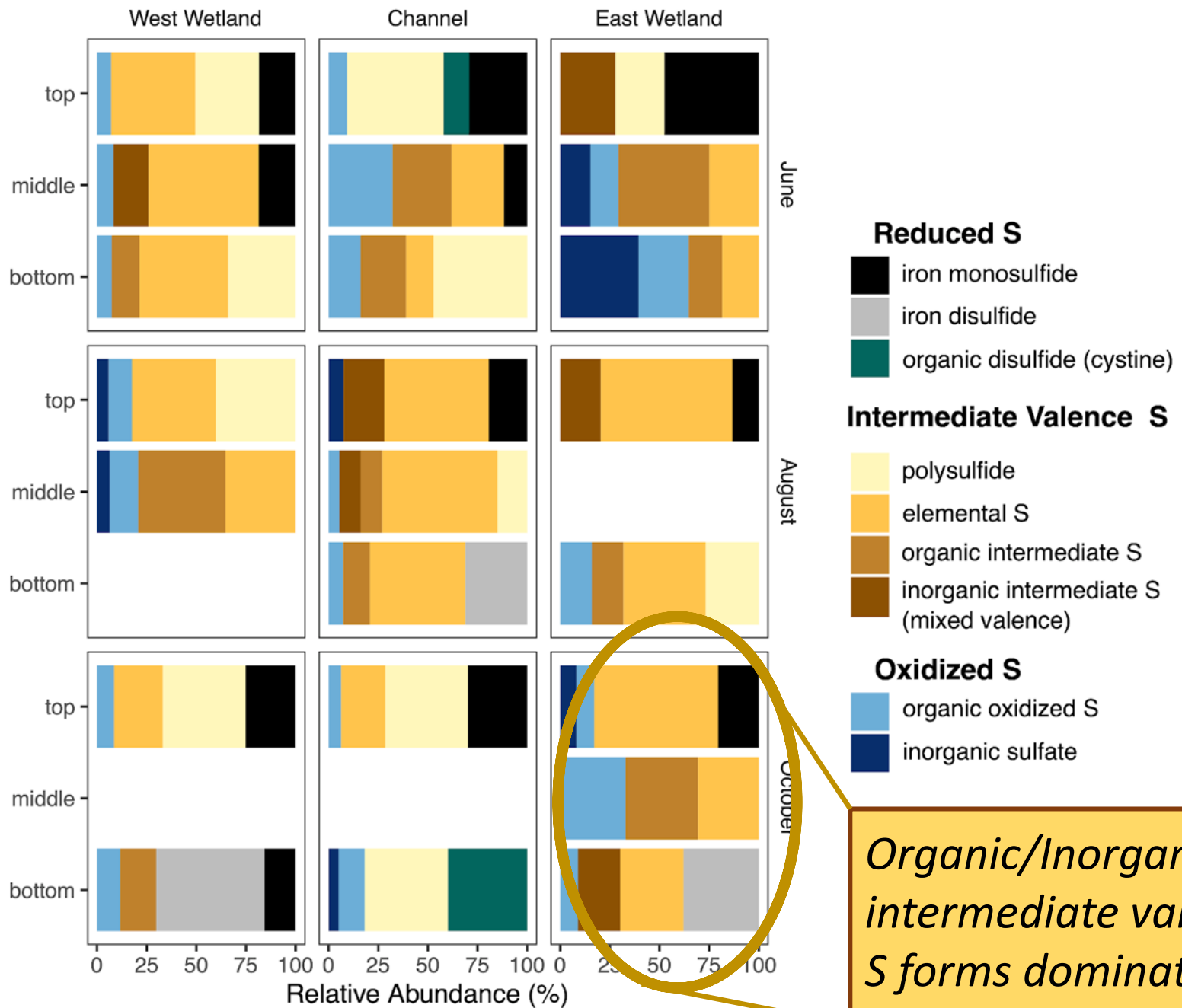


*Average sulfide consistently low ~3 μM*

- May
- June
- July
- August
- October

*Sulfide not impacted by hyporheic flux variability*

# Sediment Sulfur (S K-edge EXAFS)



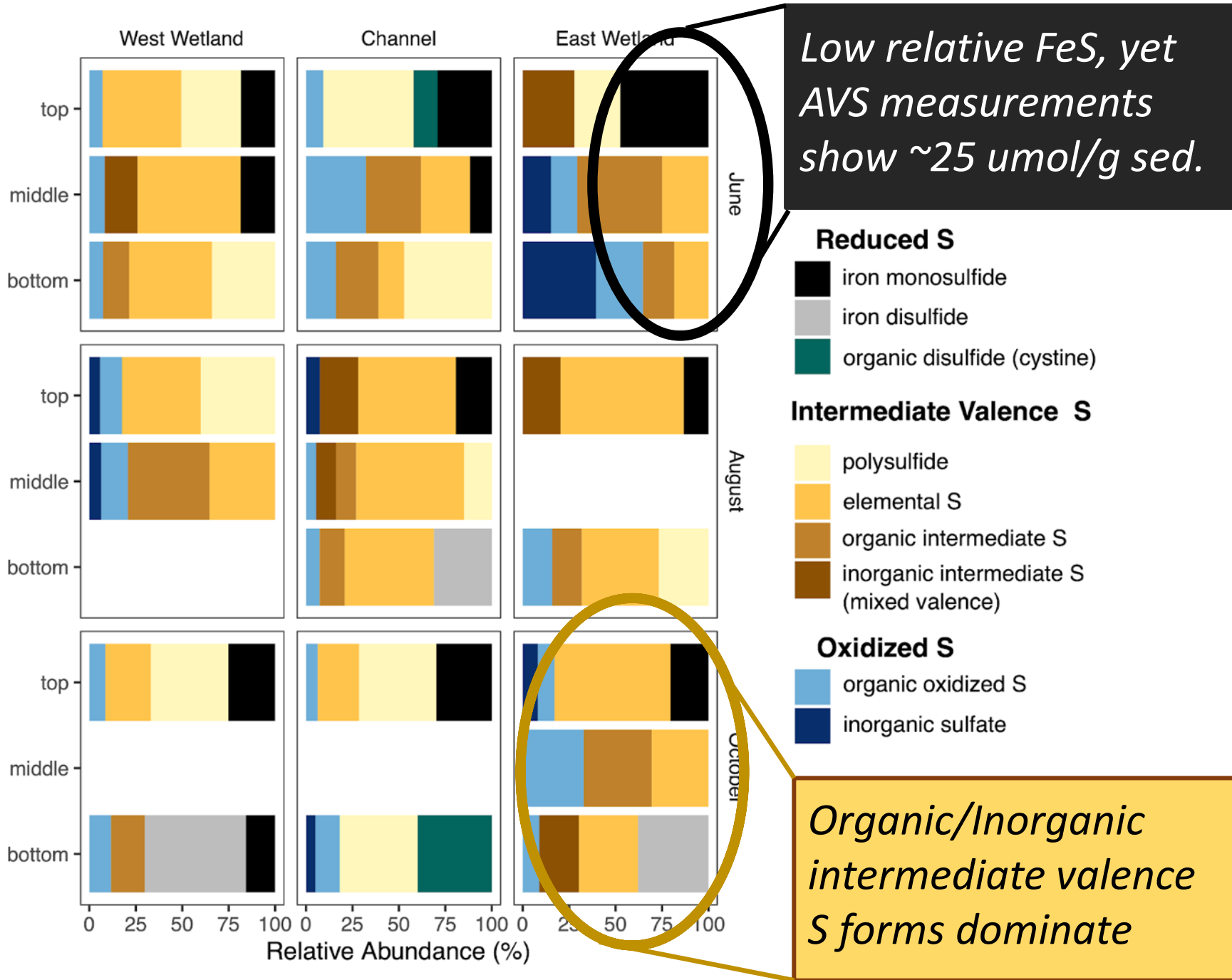
Intermediate valence S forms (polysulfide, S(0), etc.) dominate total S pool!!!

*Organic/Inorganic intermediate valence S forms dominate*

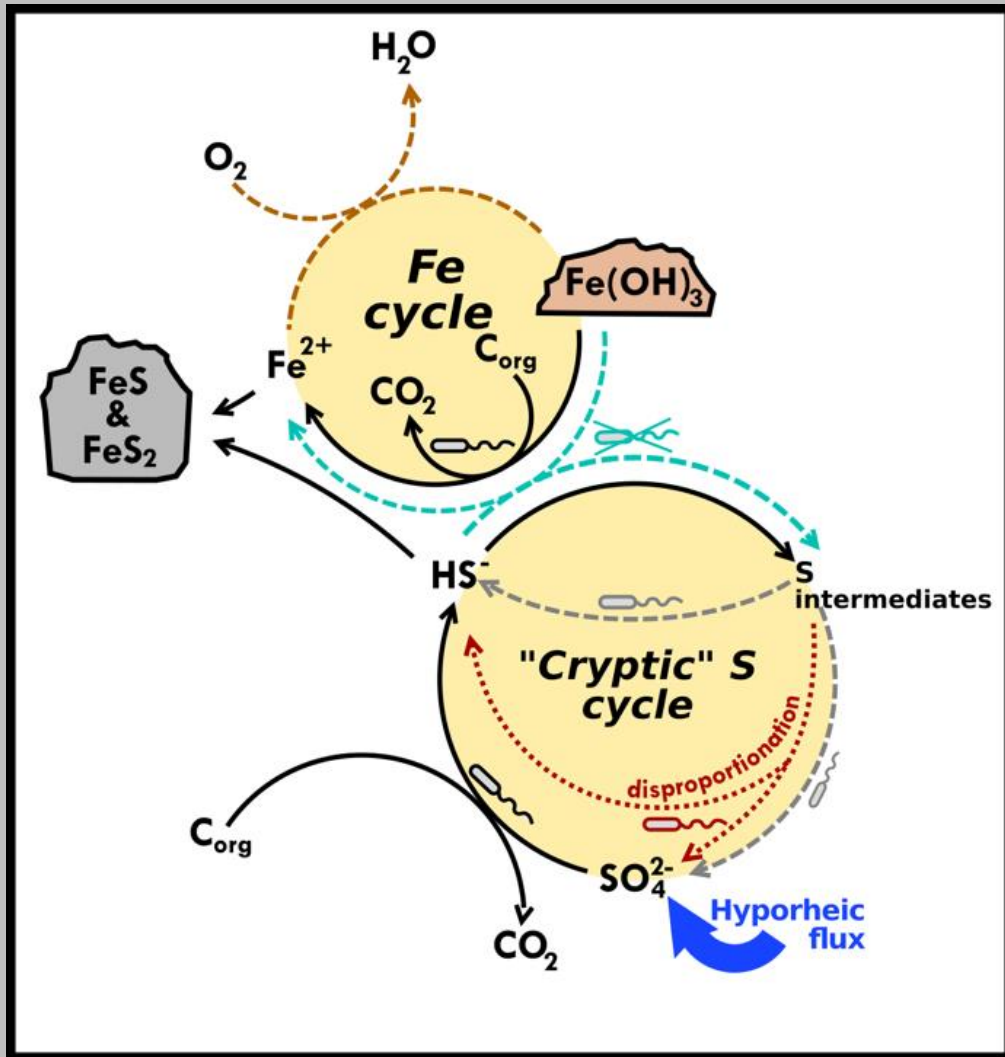
# Sediment Sulfur

(S K-edge EXAFS)

No obvious trends in S composition, but slightly higher oxidized S in October and June in East Wetland when Fe(III) oxides more abundant



Abundant S intermediates suggests active cryptic S cycle linked to Fe cycle



Sulfate from hyporheic flux or regeneration at depth from oxidation of S intermediates

*What is the fate of all these S intermediates? Are they reactive?*

*Rates of S cycling and formation/transformation of various products? How does sulfate concentration impact this?*

# Influence of dynamic hydrologic conditions

- Response and recovery rates still need attention
- Modeling should give us greater insight
- Potential implications for methane fluxes

